

EJC2017

New views on radioactivity

Les Issambres, France

24-29/09/2017

DATING METHODS IN PREHISTORY DURING QUATERNARY PERIOD

Christophe Falguères

UMR7194 "Histoire naturelle de l'Homme préhistorique"

Muséum national d'histoire naturelle

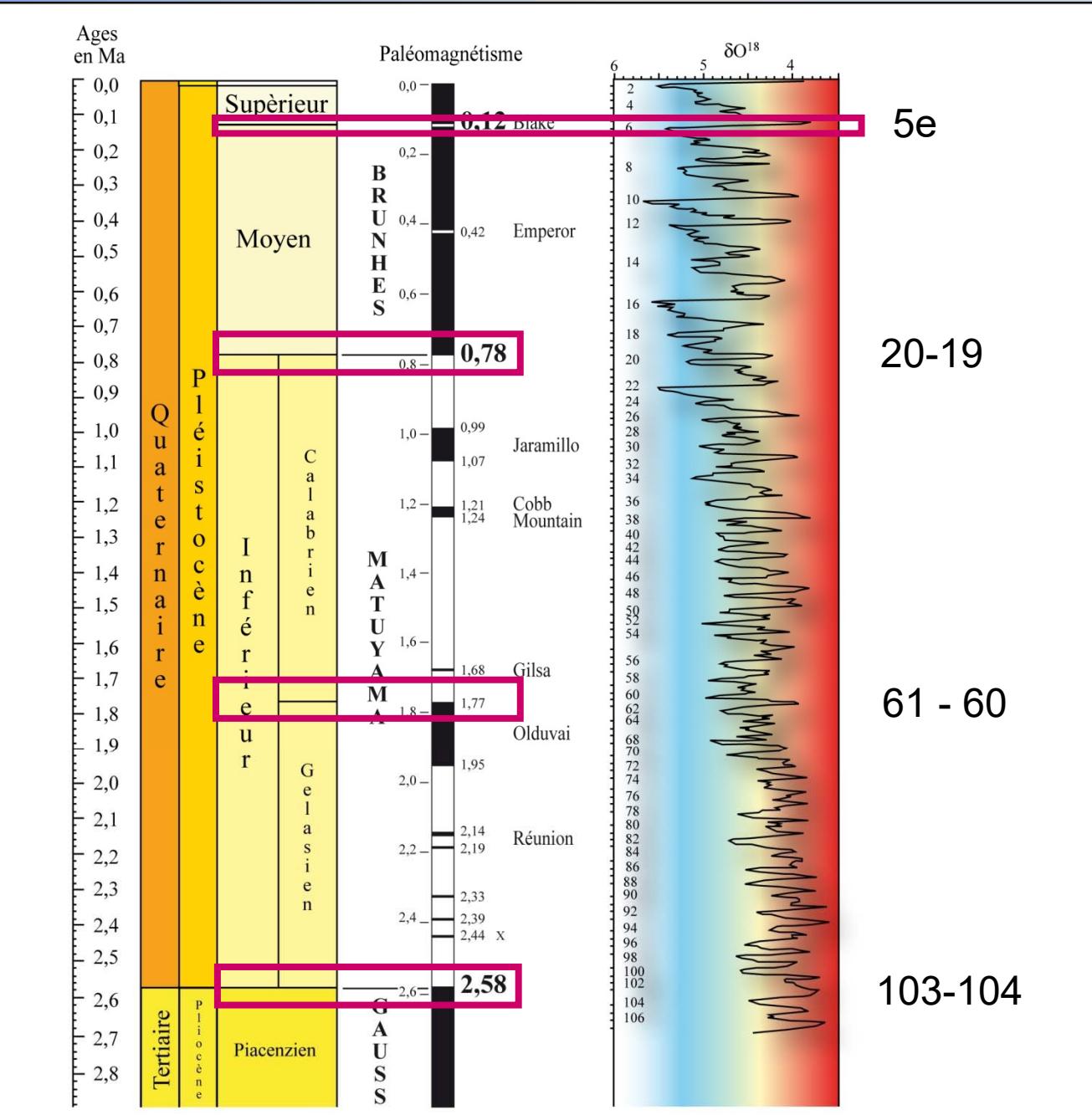
Institut de Paléontologie Humaine,

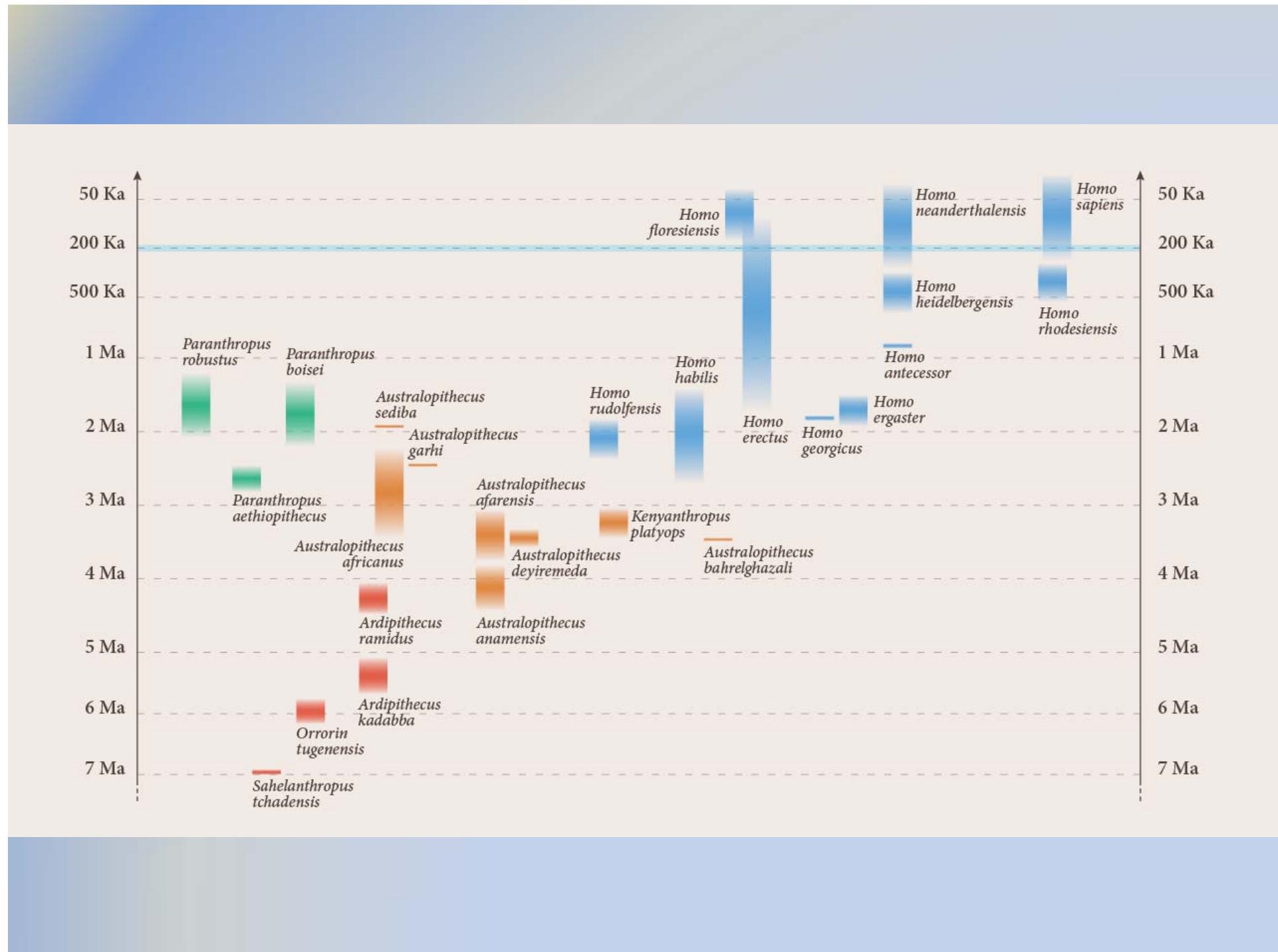
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Principales méthodes utilisées en Géochronologie

Méthodes « Naturalistes »

- Stratigraphie
- Biostratigraphie
- Varves
- Dendrochronologie
- Téphrochronologie
- Paléomagnétisme
- Isotopes de l'oxygène

Méthodes fondées sur la croissance ou la décroissance radioactive

- Radiocarbone, ^{14}C
- Famille de l'Argon, Ar/Ar, K/Ar
- Séries de l'uranium, Th/U, Pa/U
- Autres cosmonucléides, Al/Be

Méthodes fondées sur un phénomène chimique

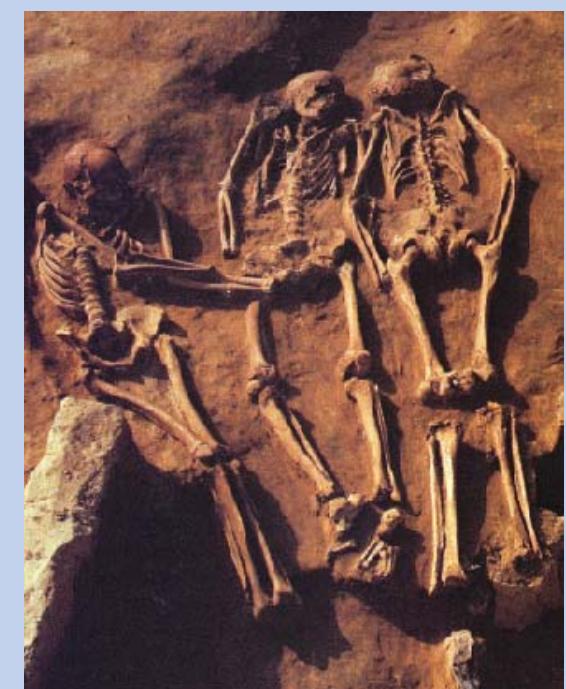
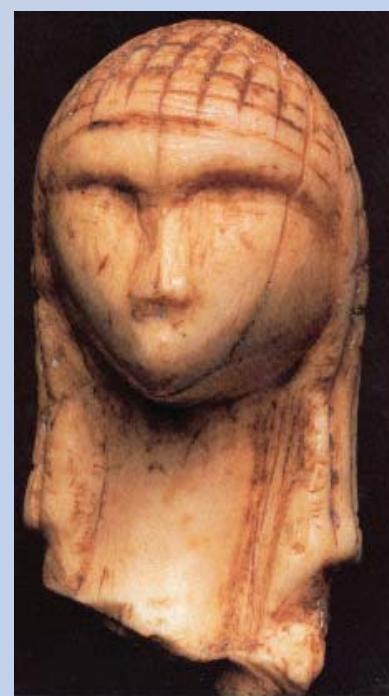
- Racémisation des acides aminés
- Hydratation de l'obsidienne

Méthodes fondées sur les dommages créés dans les minéraux par la radioactivité naturelle

- Trace de fission
- Résonance de Spin Électronique, ESR
- Luminescence, TL, OSL, TT-OSL, IRSL



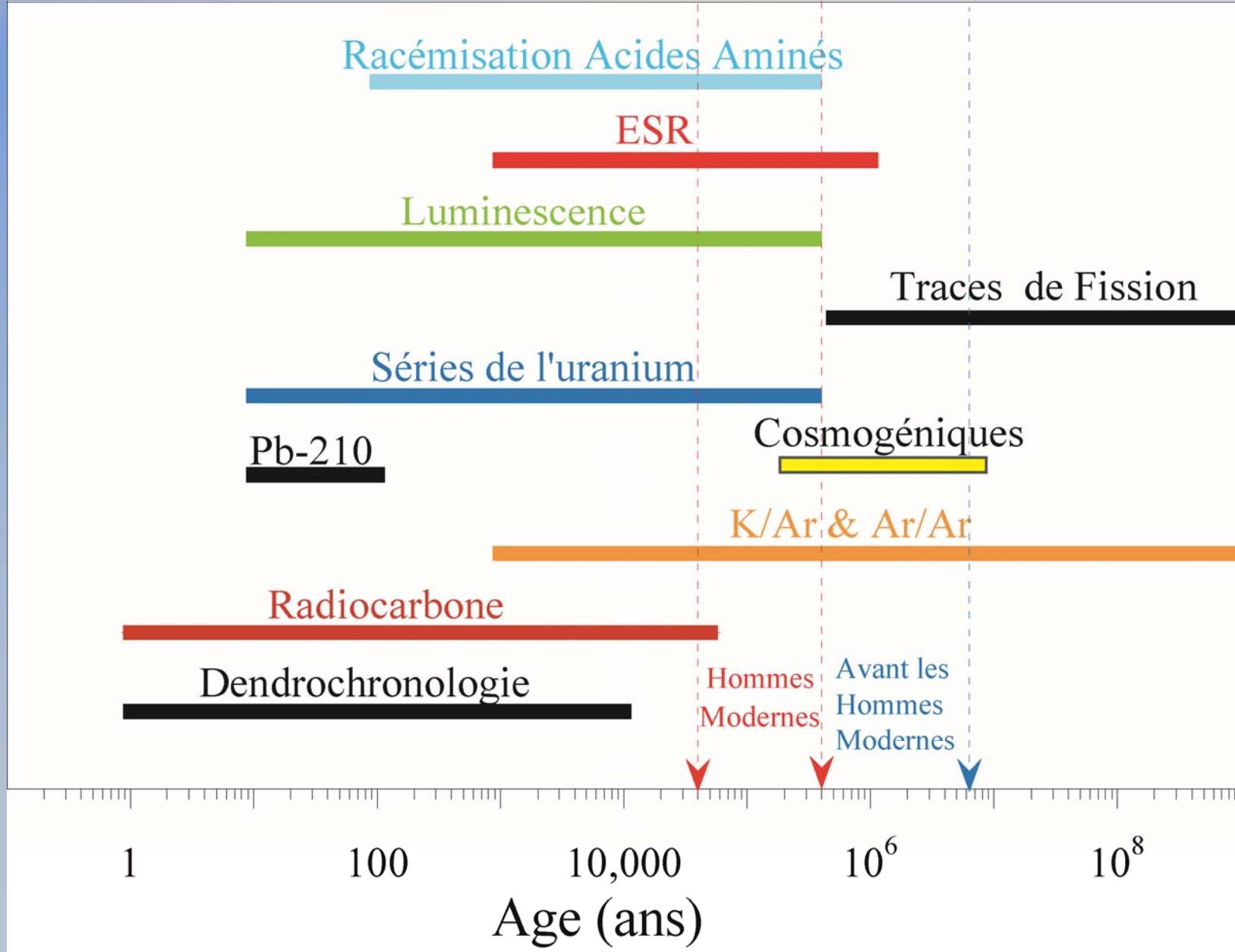
What we want to date...





... and what we really date...





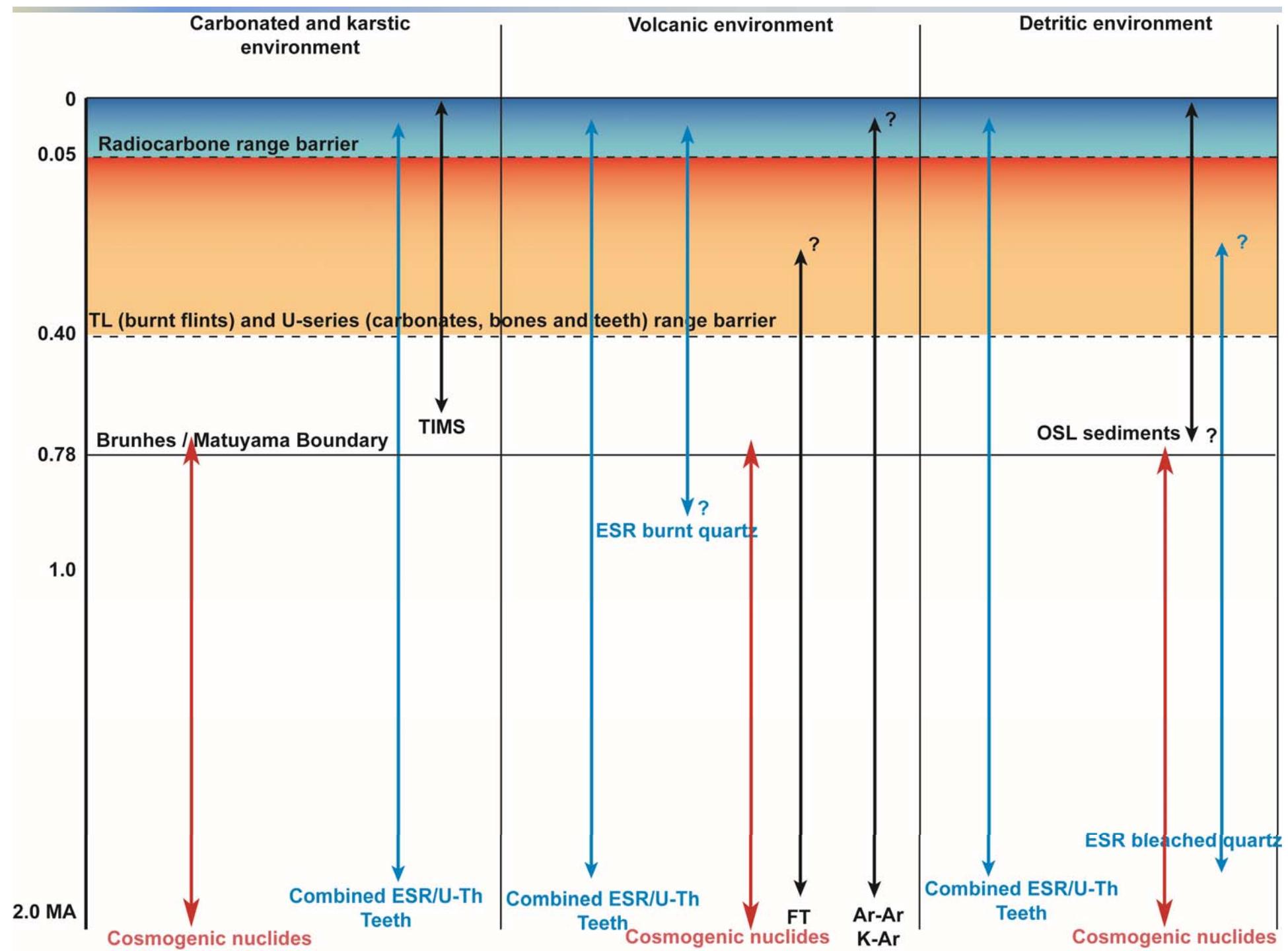
Problems related to the dating methods

Range of applicability

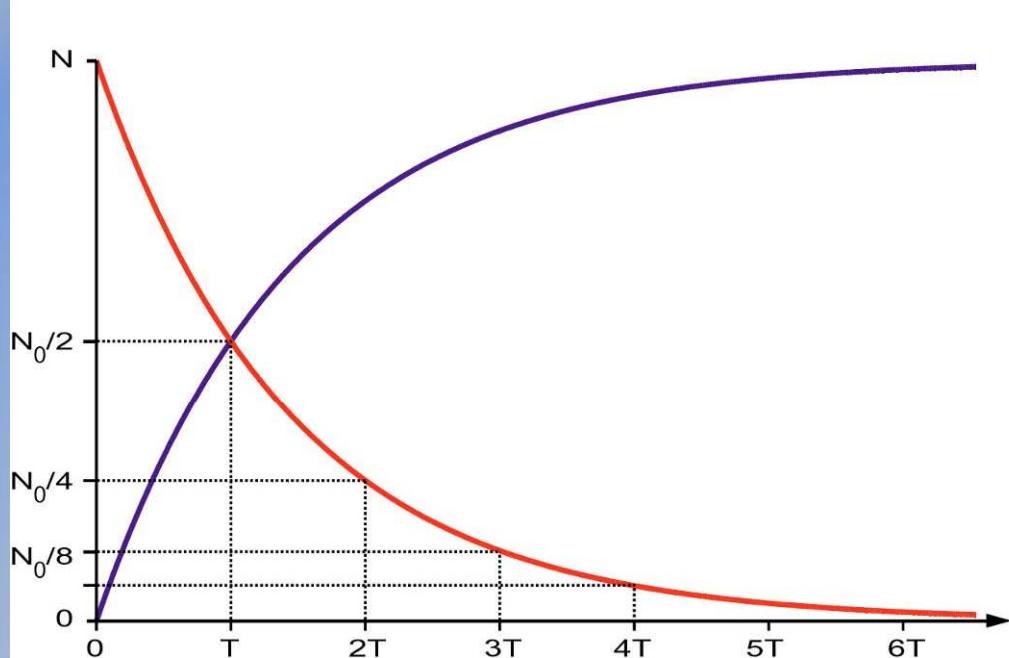
Uranium uptake => closed system or opened system

Suitability of the samples

Stratigraphic control and heterogeneity of the layers



Isotopic dating methods



$$N(t) = N_0 (1 - e^{-0.693t/T})$$

Elément fils

$$N(t) = N_0 e^{-\lambda t}$$

$$N(t) = N_0 e^{-0.693t/T}$$

Elément père

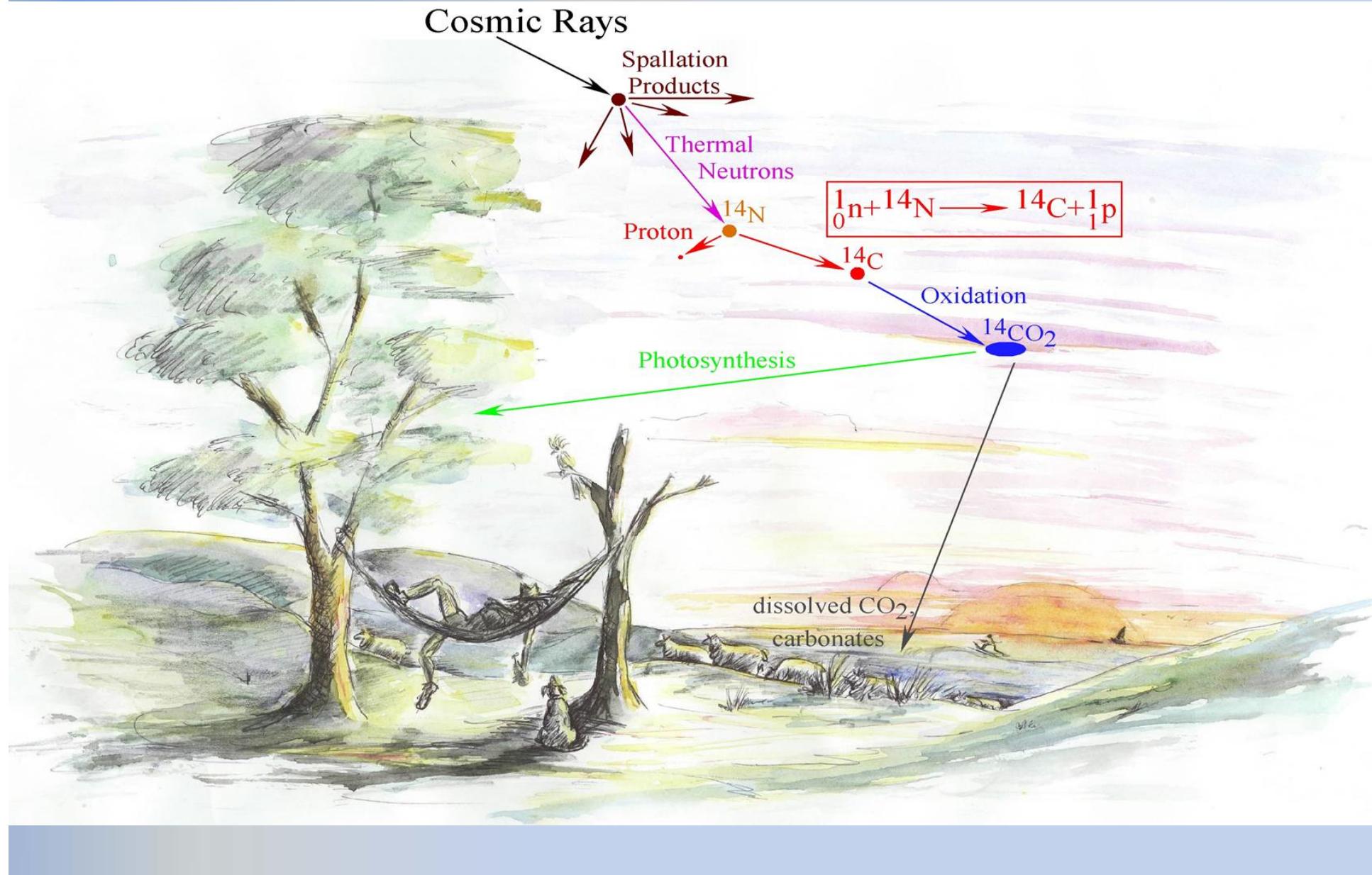
Half-life

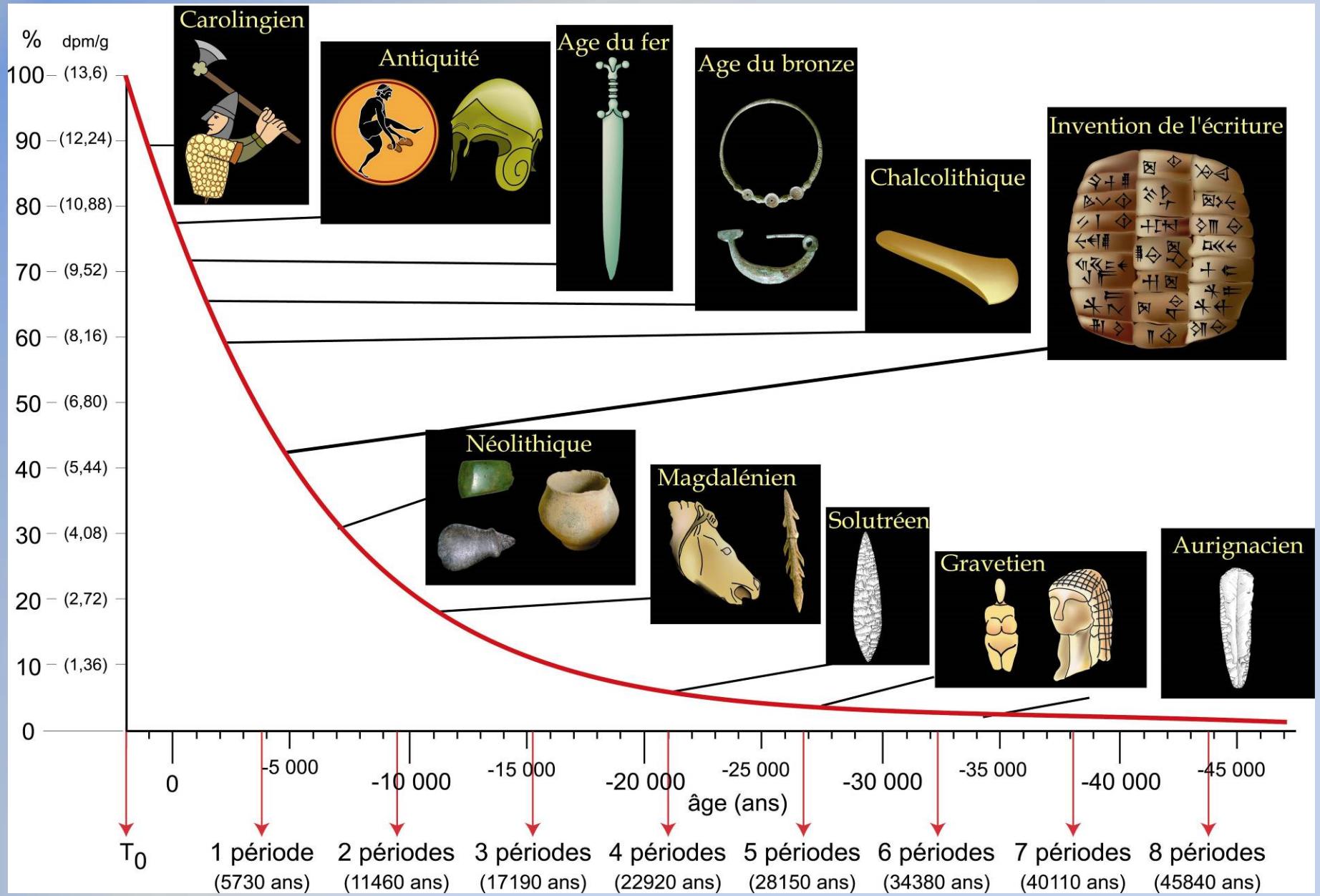
^{14}C : 5,730 ka

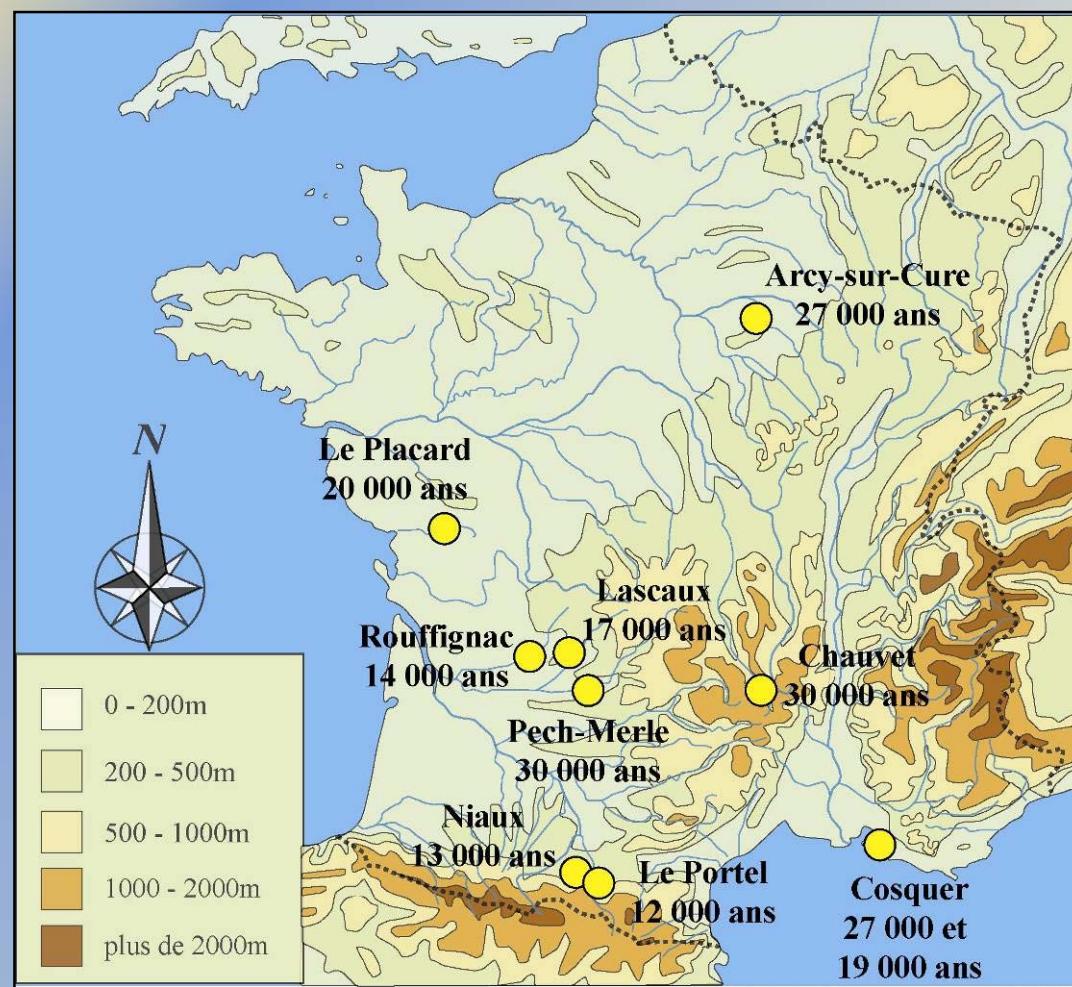
^{230}Th : 75,200 ka

^{40}K : 1 250 Ma

Radiocarbon method



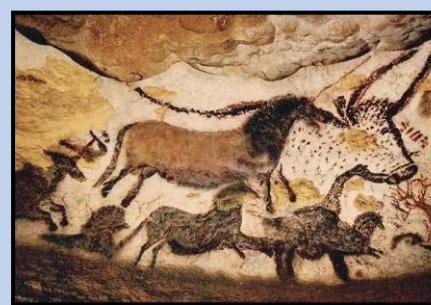




Arcy-sur-Cure



Chauvet



Lascaux



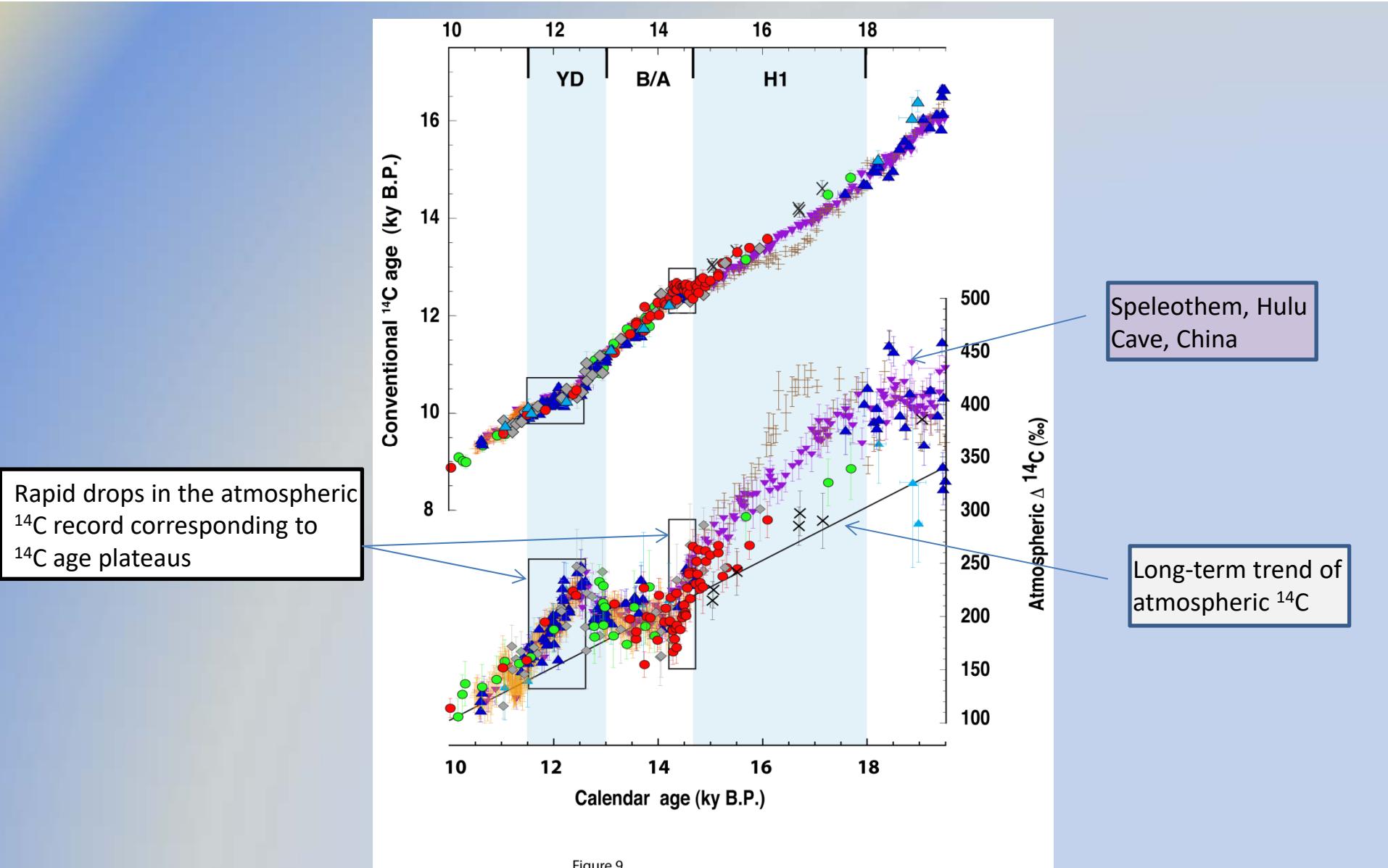
Niaux



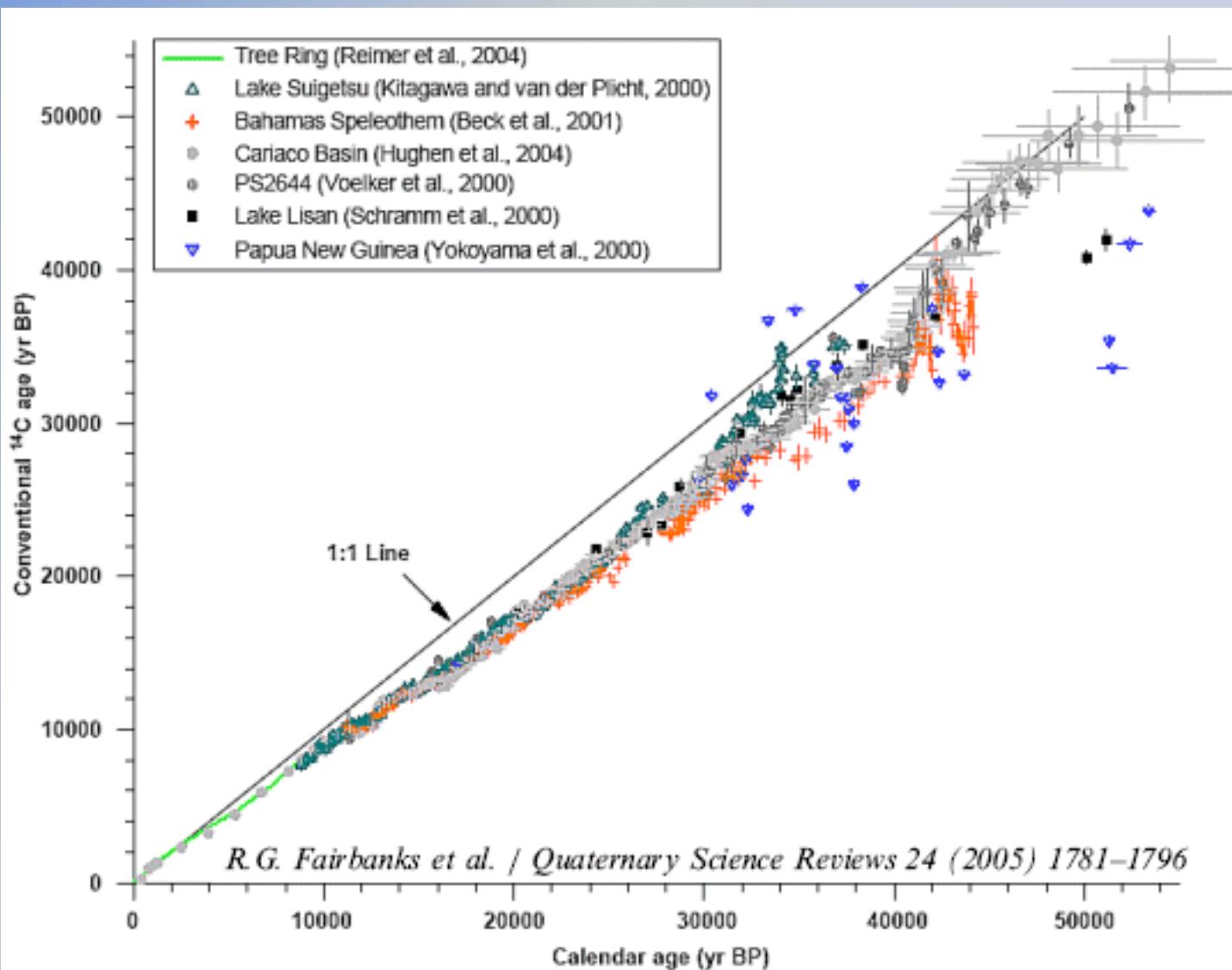
Le Portel



Cosquer



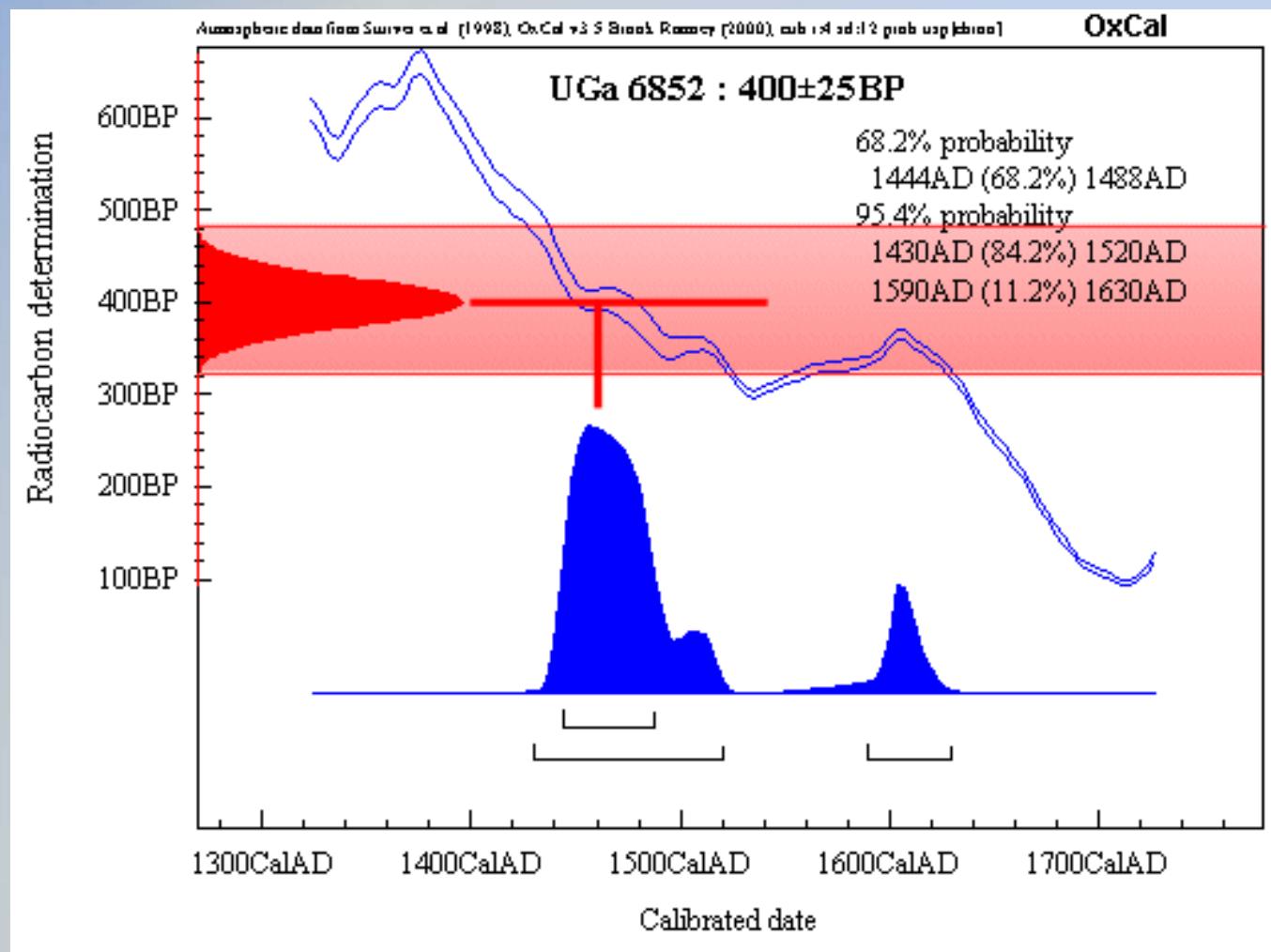
Extremely large variations of atmospheric ^{14}C concentration during the whole deglaciation period implying a correction of the ages (Durand et al., Radiocarbon, 2013)



^{14}C ages can be presented as ^{14}C BP (conventional age) based on the proportion of radiocarbon in the sample without any variation of the atmospheric radiocarbon concentration.

They can be calibrated (INTCAL13 using data sets from ^{14}C measurements on tree rings, plant macrofossils, speleothems, corals, foraminifera; **Reimer et al., Radiocarbon, 2013**)

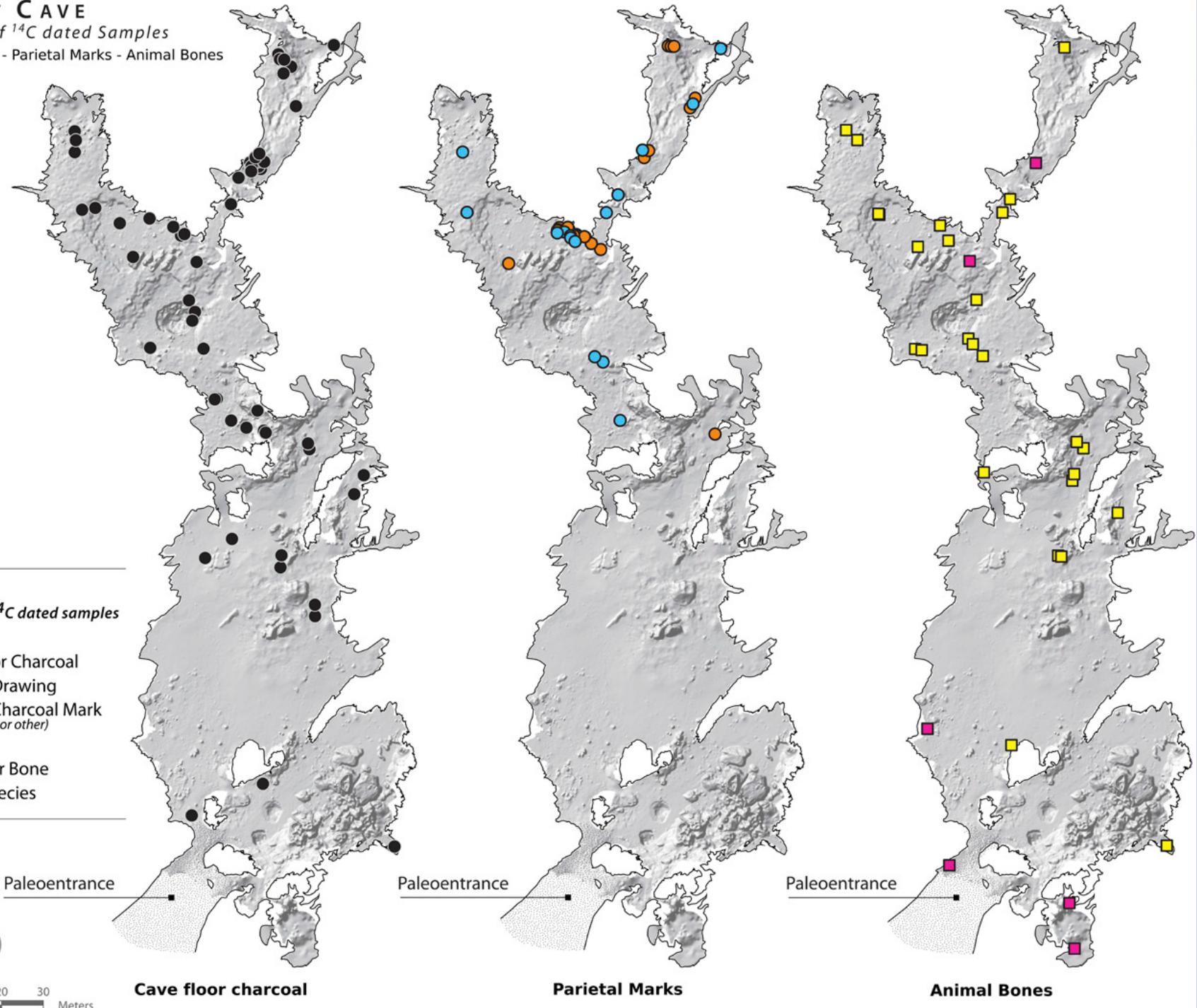
^{14}C cal. BP or directly in calendar ages BP, BC or AD

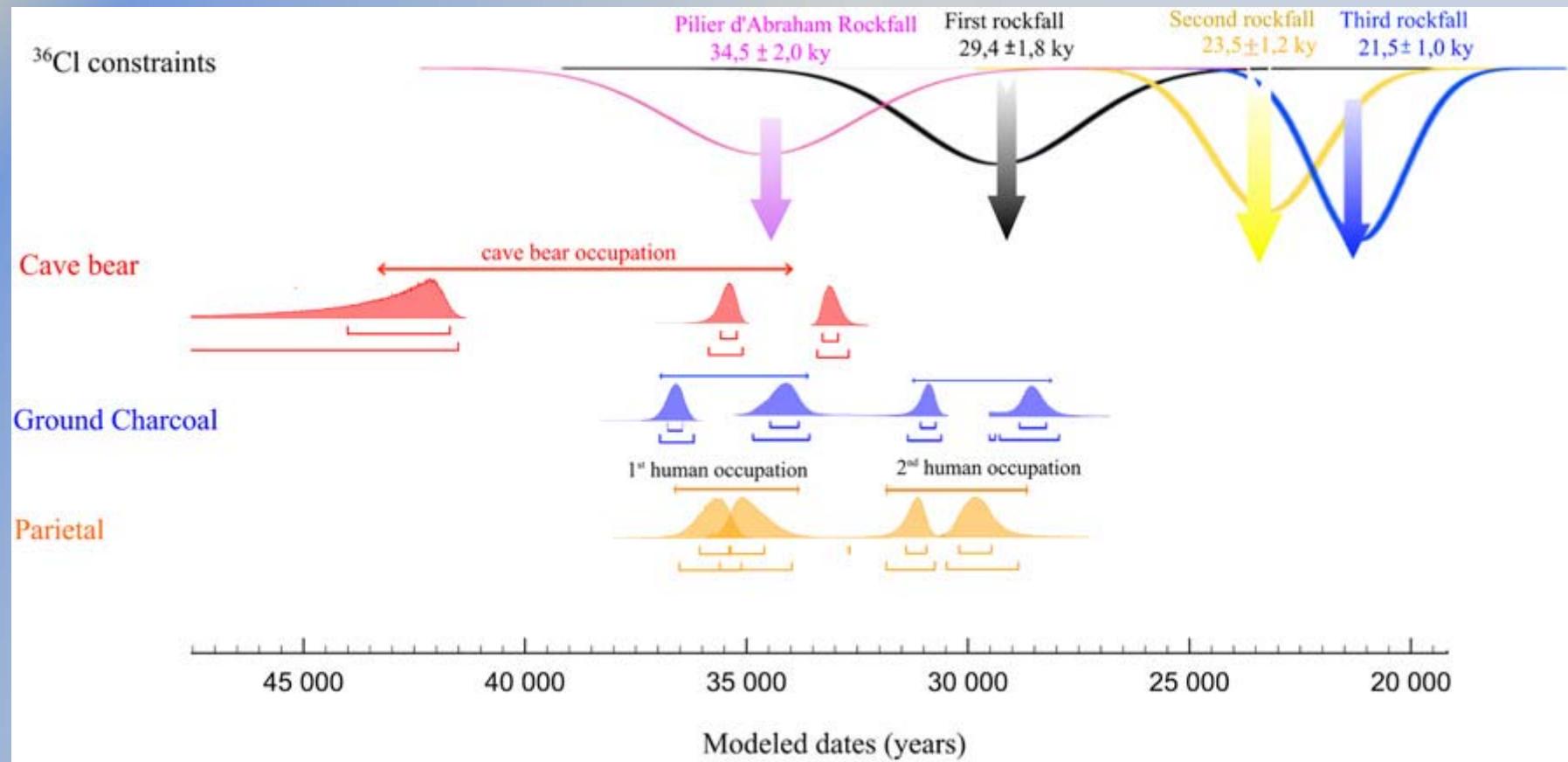


CHAUVET CAVE

Distribution of ^{14}C dated Samples

Cave floor charcoal - Parietal Marks - Animal Bones

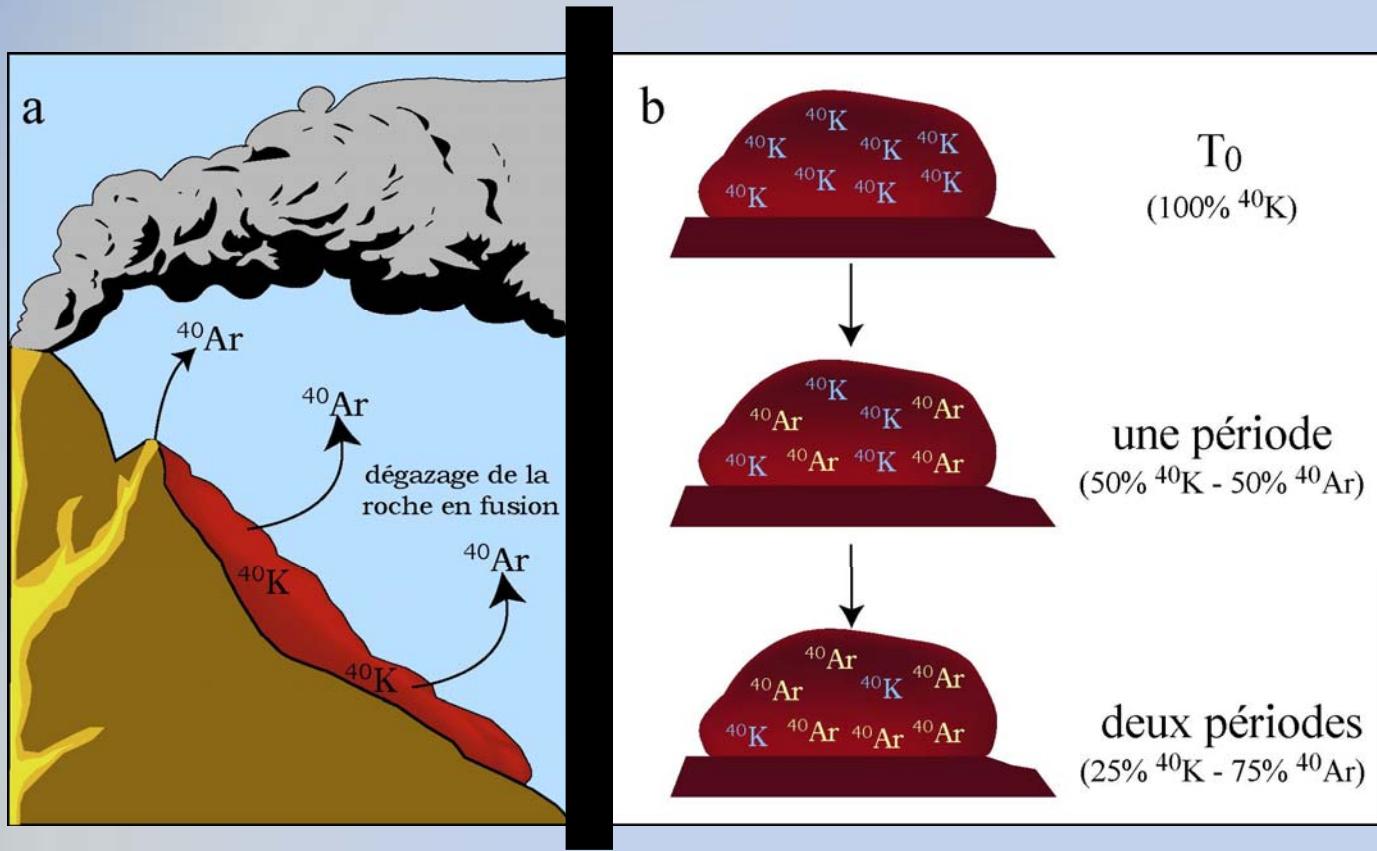
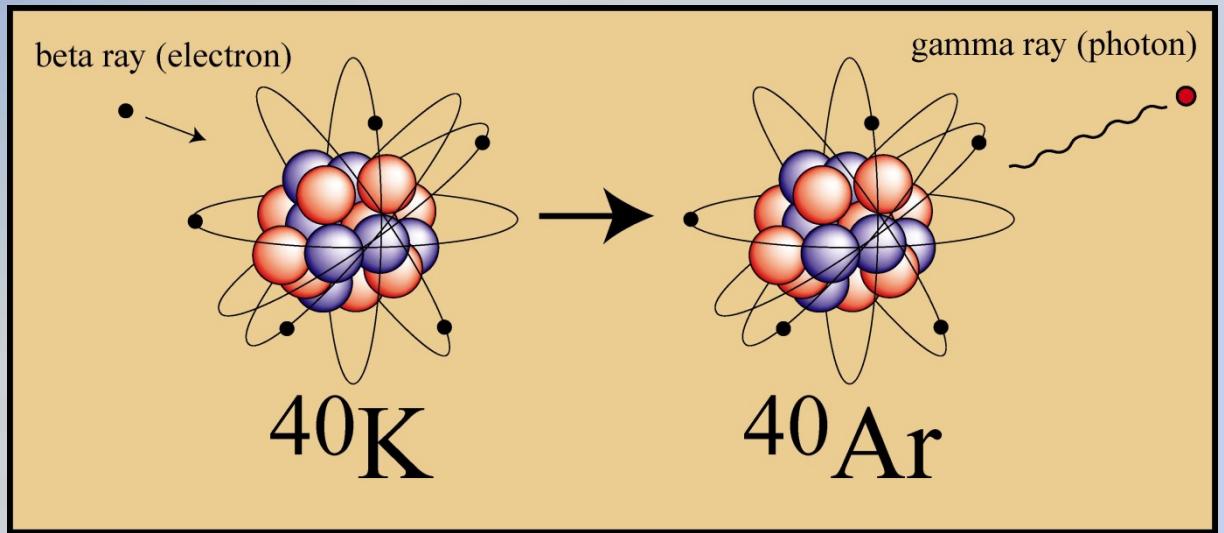




High-precision Bayesian model obtained for the Chauvet-Pont d'Arc Cave. Modeled boundaries for the start and end of each occupation phase are represented in red for the Cave Bear model (postulating a continuous occupation), in blue for the Cave Floor Charcoal model, and in orange for the Parietal model. Two distinct human occupations are clearly identified, extending from 37,000 to 33,500 y ago for the first one, and from 31,000 to 28,000 y ago for the second one. Cave bear presence in the cave is attested until 33,000 y ago. (Quiles et al., PNAS, 2016)

Argon methods

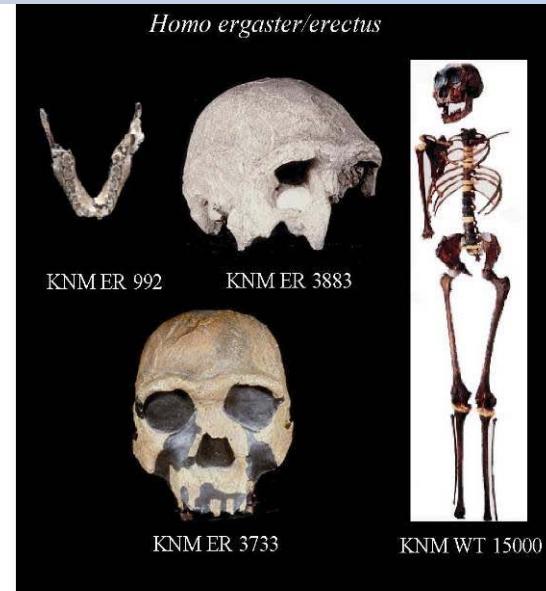
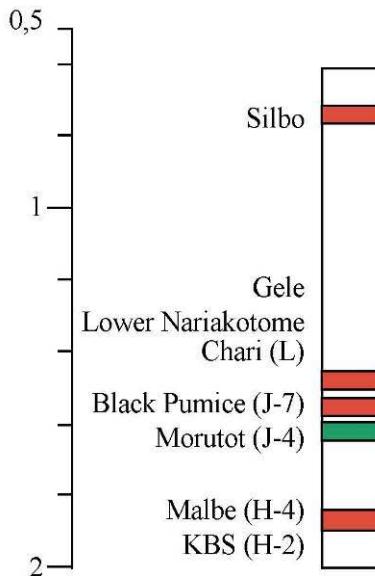




East Africa

Ages de quelques niveaux
volcaniques
du Bassin de l'Omo-Turkana
(d'après Brown, 1994)

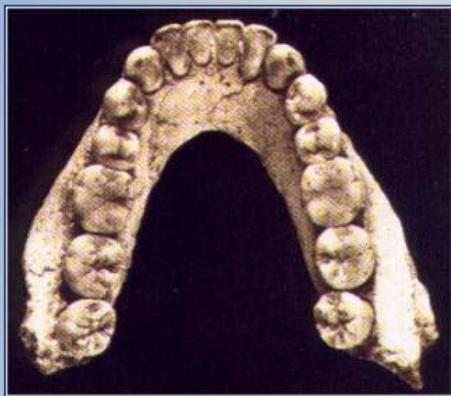
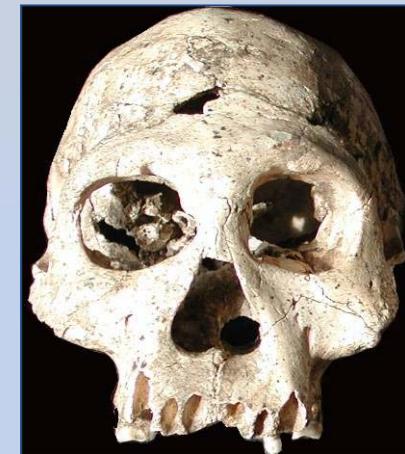
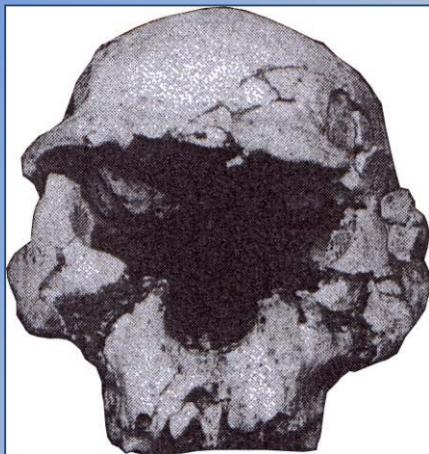
Age (Ma)



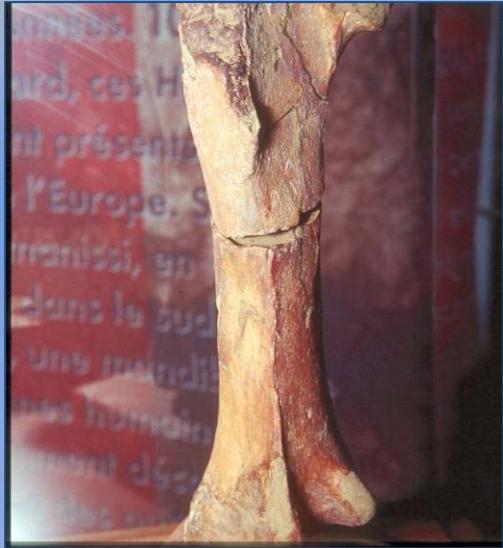


Dmanisi site, Georgia

The first settlement out of Africa



5 skulls and 4 mandibles + post-cranial human remains
associated to a Lower Pleistocene fauna and artefacts>



Struthio dmanisensis



Megantereon megantereon



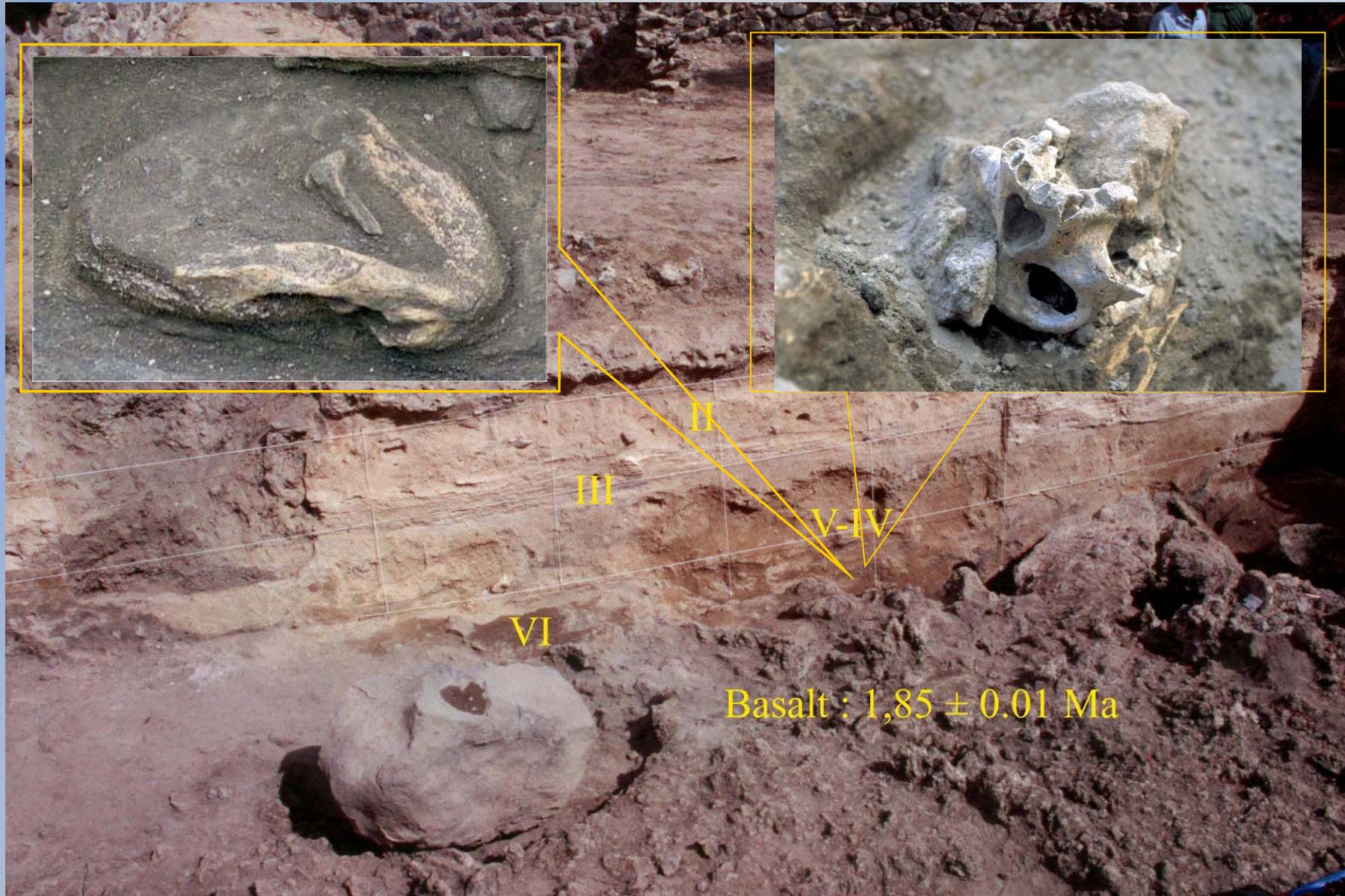
Archidiskodon meridionalis



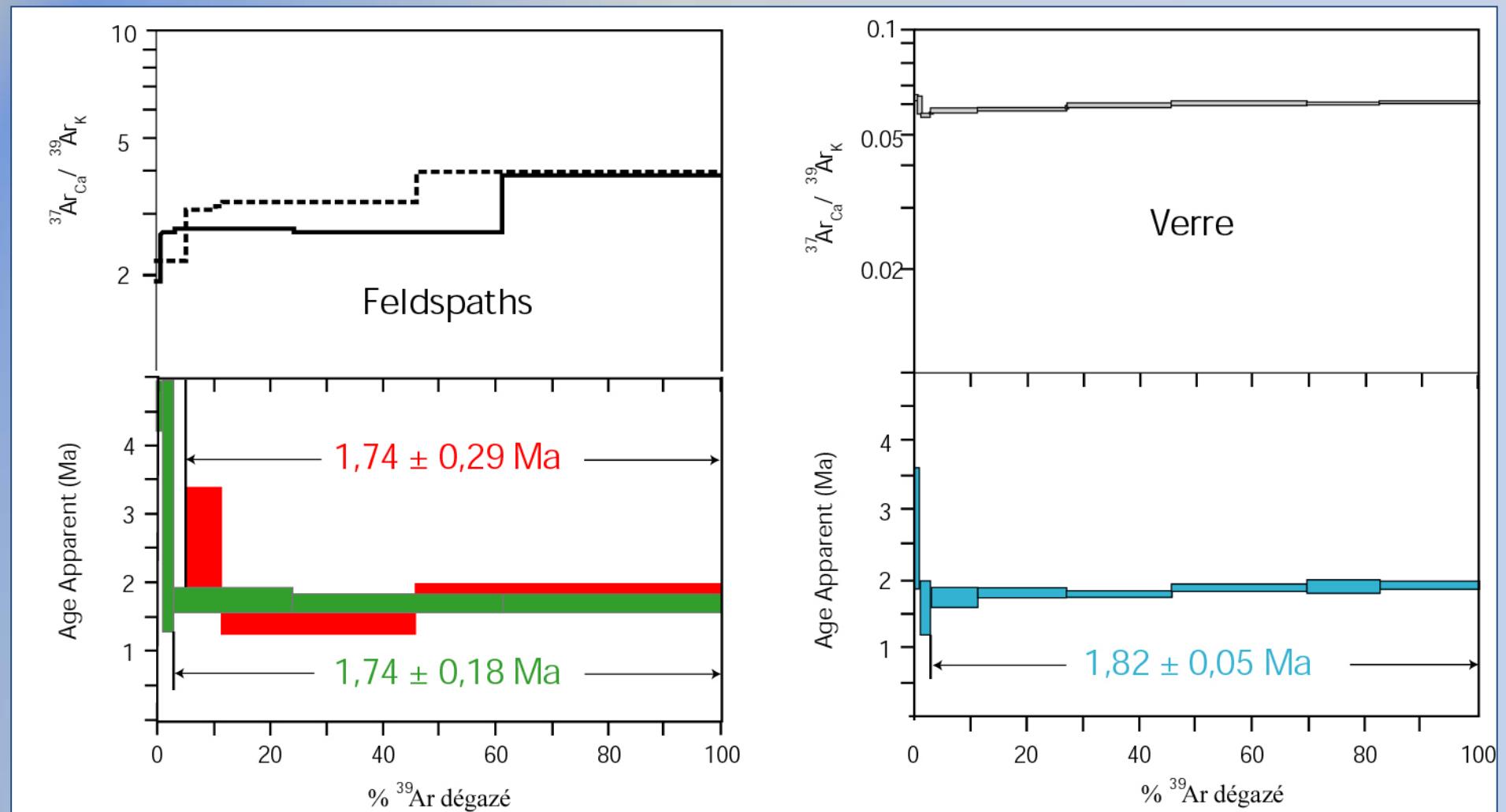
Chopping Tool



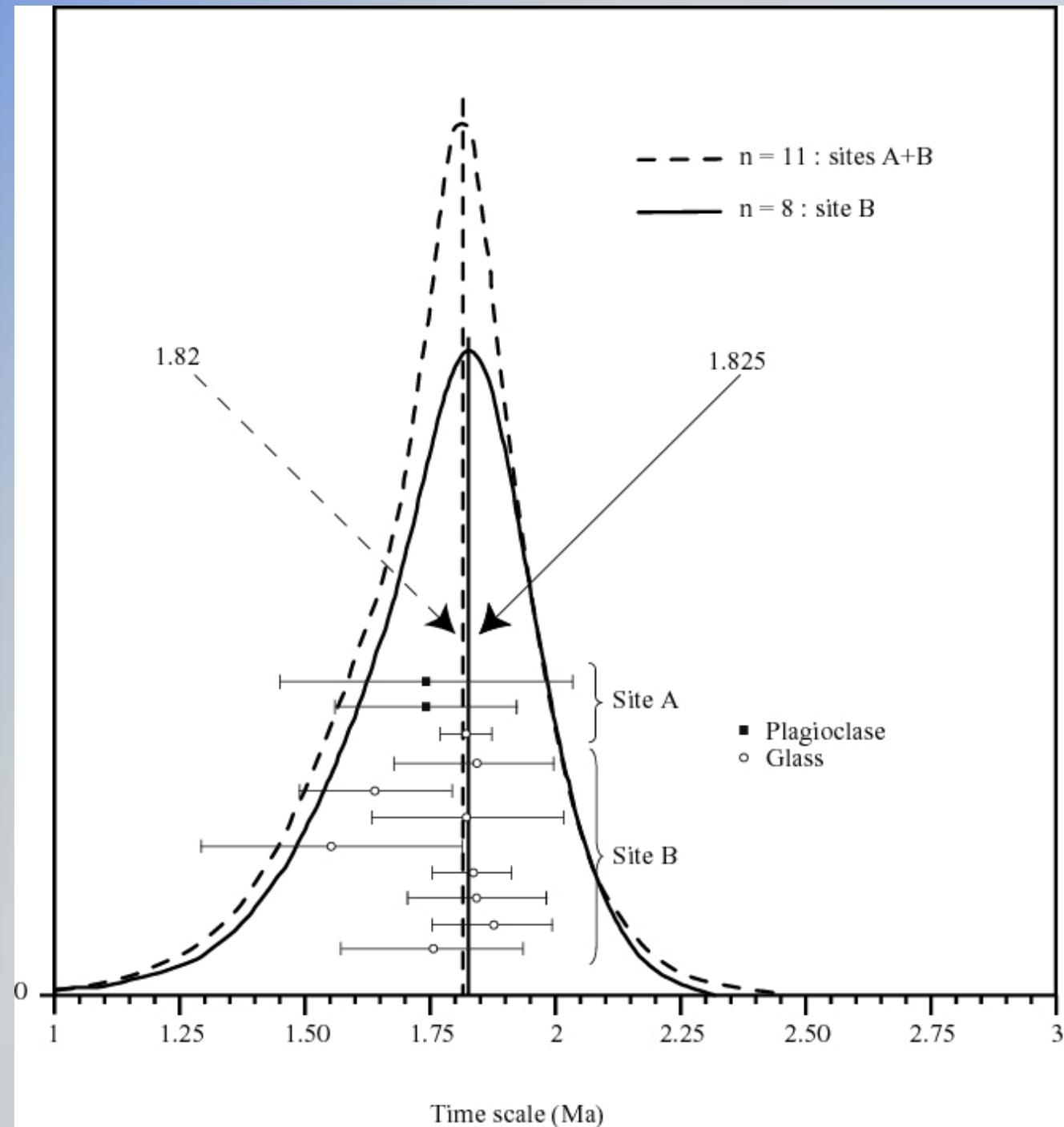
Chopper

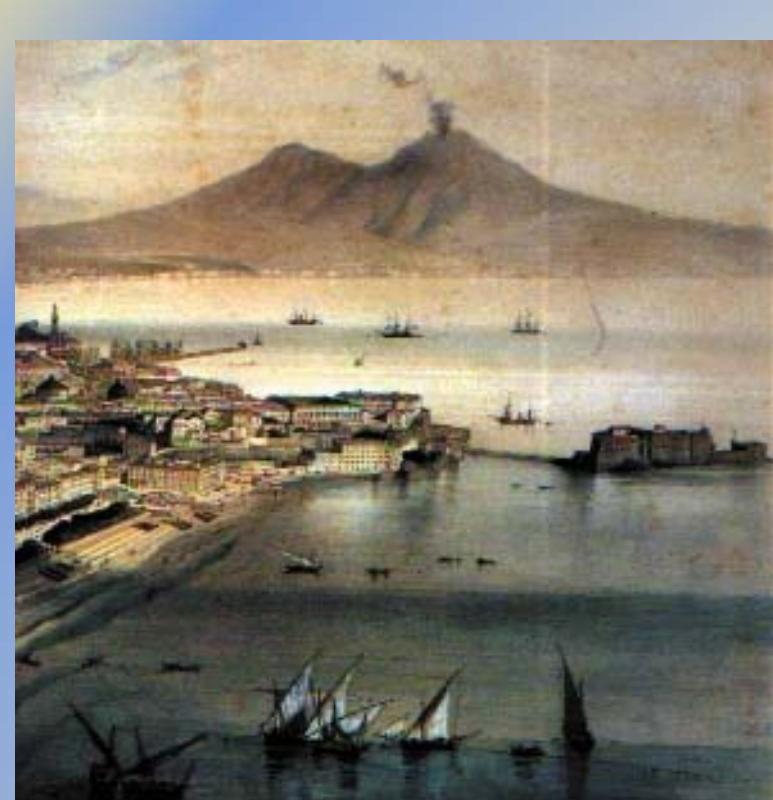


$^{40}\text{Ar}/^{39}\text{Ar}$ Age spectra on the D2600 mandible level



Weighted Mean : $1,81 \pm 0,05 \text{ Ma}$
 (Garcia et al., Quat Geochrono, 2010)





POMPEI et HERCULANUM, 79 AD

ISOCHRON $^{40}\text{Ar}/^{39}\text{Ar}$ AGE

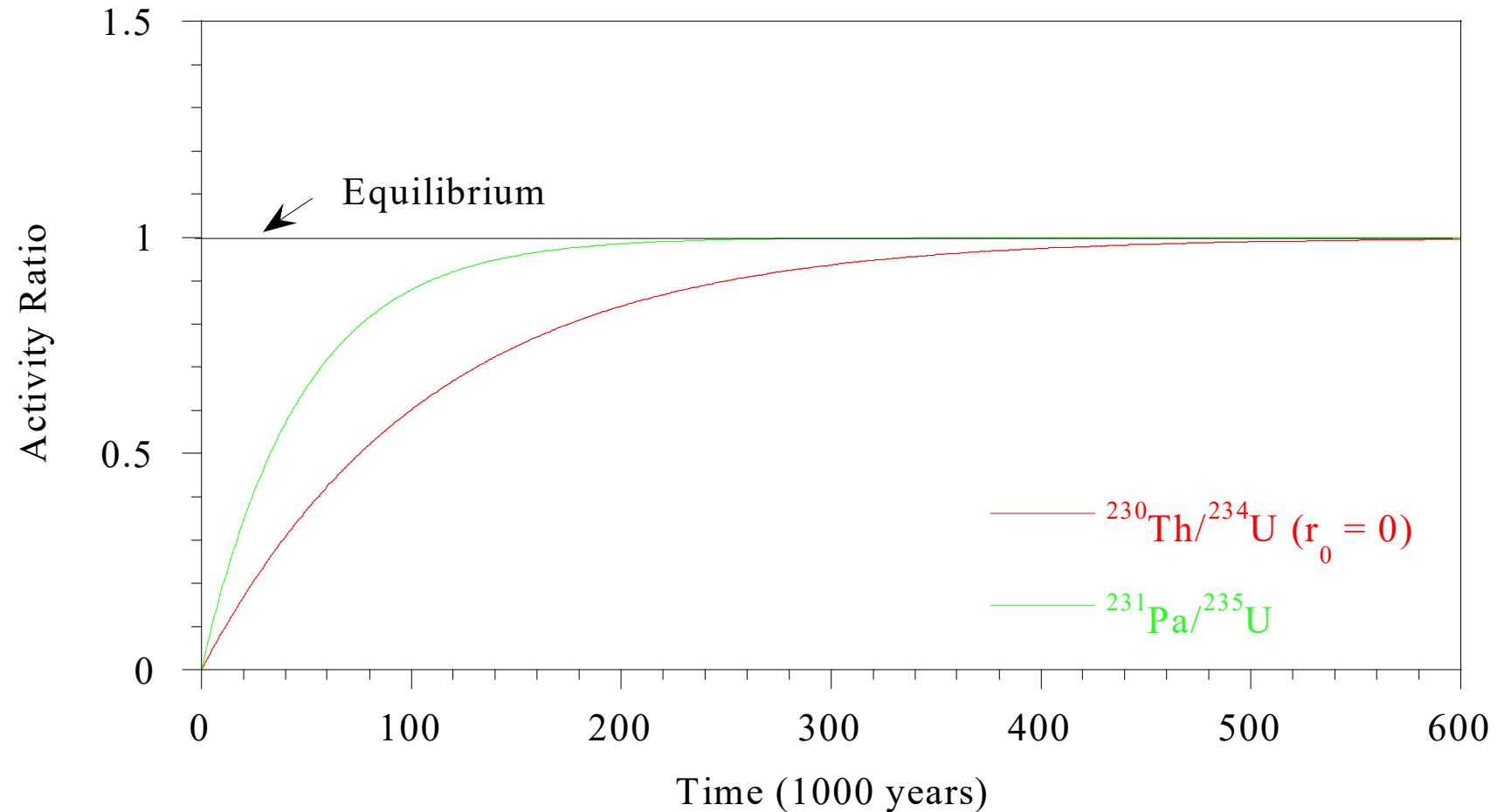
1925 ± 94 years (2 sigma error)



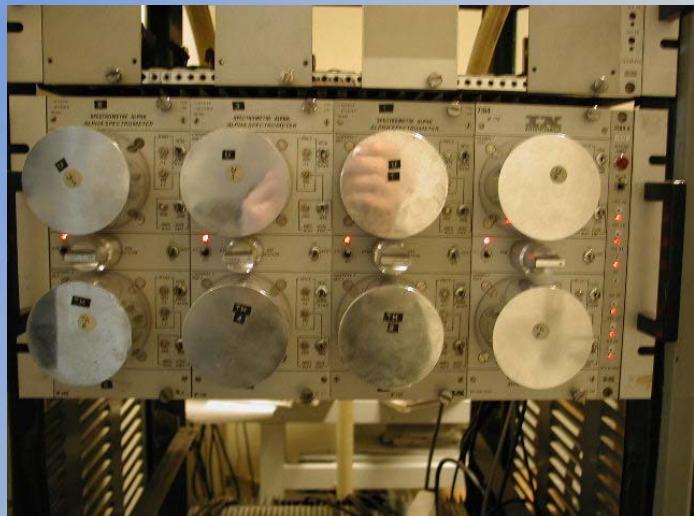
(Renne et al., *Science*, 1997)

Uranium-series methods

	238-Uranium Decay Series	232-Thorium Decay Series	235-Uranium Decay Series
U	^{238}U $4,47 \cdot 10^9$ a	^{234}U $2,45 \cdot 10^5$ a	^{235}U $7 \cdot 10^8$ a
Pa	^{234}Pa 6,69 h	: alpha decay	^{231}Pa $3,3 \cdot 10^4$ a
Th	^{234}Th 24,1 d	^{230}Th $7,5 \cdot 10^4$ a	^{232}Th $1,4 \cdot 10^8$ a
Ac		^{228}Ac 6,15 h	^{228}Th 1,91 a
Ra	^{226}Ra $1,6 \cdot 10^3$ a	^{228}Ra 5,75 a	^{224}Ra 3,66 d
Rn	^{222}Rn 3,823 d	^{220}Rn 55,6 s	^{219}Rn 3,96 s
Po	^{218}Po 3,04 m	^{216}Po 0,15 s	^{215}Po 1,78 ms
Bi	^{214}Bi 19,7 m	^{212}Bi 1,01 h	^{211}Bi 2,14 m
Pb	^{214}Pb 26,9 m	^{212}Pb 10,6 h	^{211}Pb 36,1 m
Tl	^{210}Pb 22,6 a	^{208}Tl 3,05 m	^{207}Tl 4,77 m
	^{206}Pb Stable	^{208}Pb Stable	^{207}Pb Stable



Activity measurement by alpha, gamma counting



$A_{238}, A_{234}, A_{230}, A_{232}$
 $[U] = 0.3 \mu\text{g/g} \rightarrow$
0.2 decay/mn



TIMS

Or measurement of N by
mass spectrometry
Atom measurement/second
Using the decay law

$$A = N * \lambda$$
$$[U] = 0.3 \mu\text{g/g} \rightarrow$$
$$^{238}\text{N} = 8 * 10^{14} \text{ atoms}$$



ICPMS-MC

Alpha Spectrometry

Accuracy ~ 1 to 10%

1 µg U => 3 -10 g of carbonate

TIMS

Accuracy ~ 0.2-0.5%

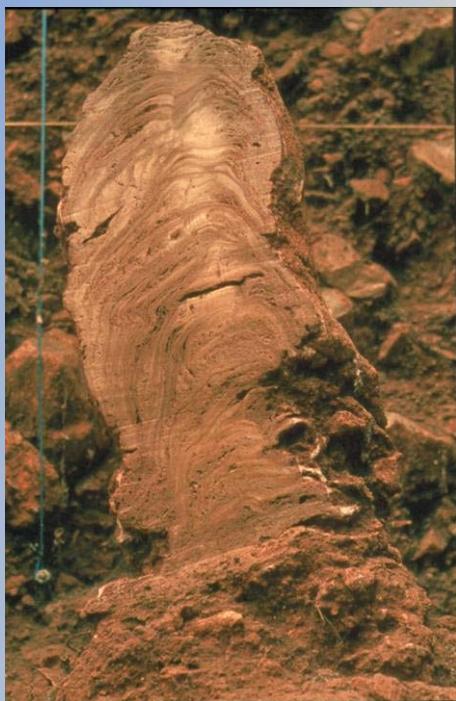
0.2 µg U => 0.5 g of carbonate

Laser Ablation – MC-ICPMS – ICP-QMS

Accuracy ~ 0.1-0.5%

0.05 -0.2 µg U => 0.1-0.5g of carbonate

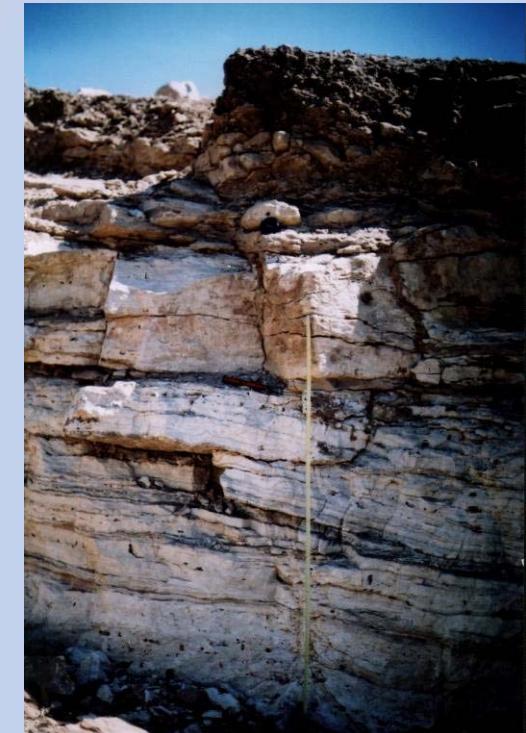
CONTINENTAL CARBONATES



Stalagmite



Stalagmitic floor

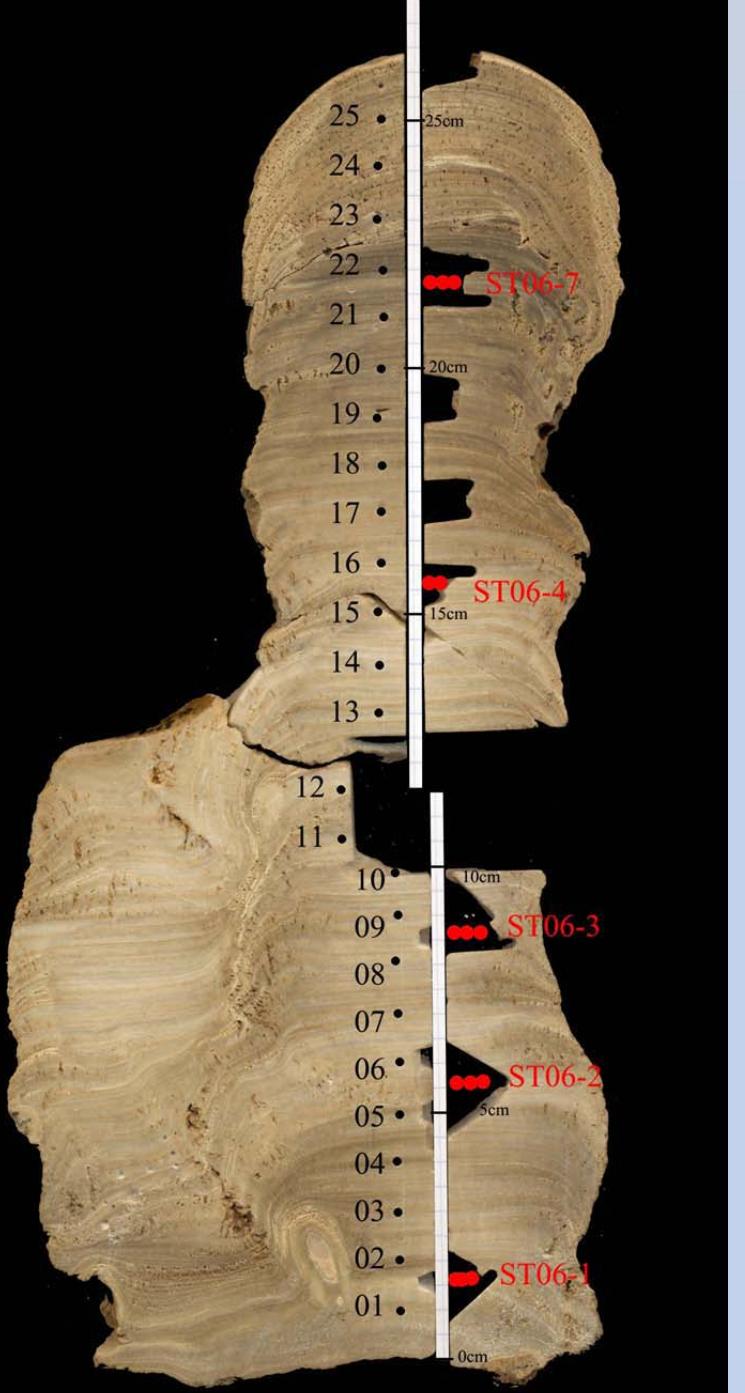
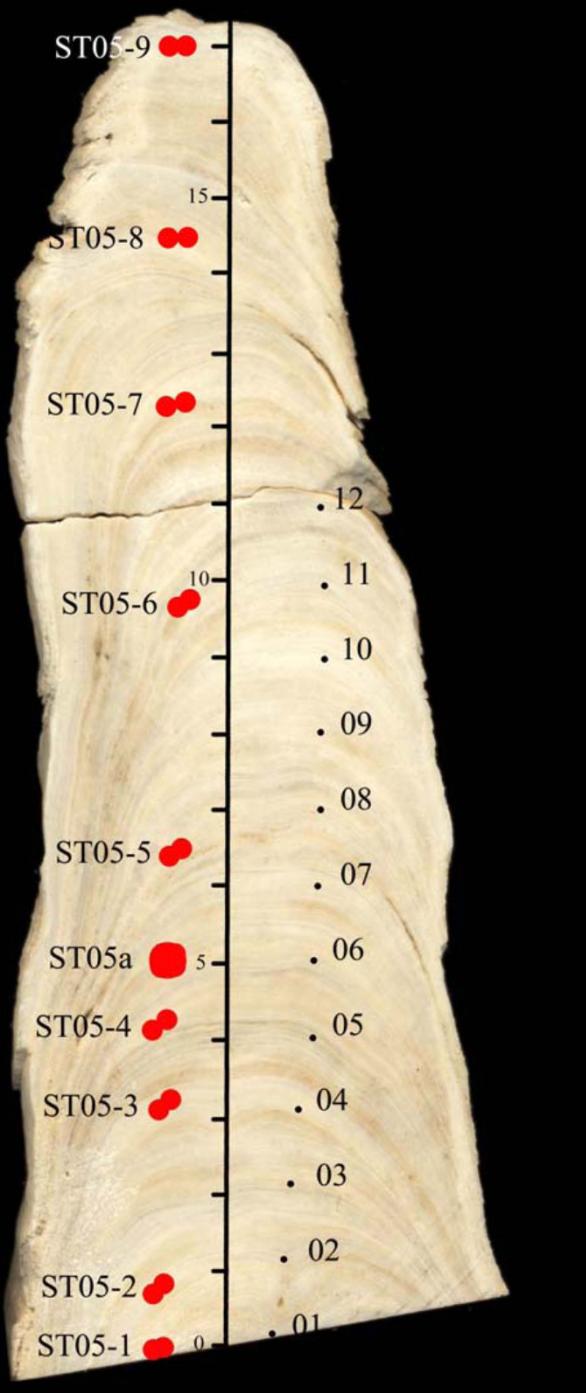


Travertines

Dating prehistoric sites in karstic area by uranium-series

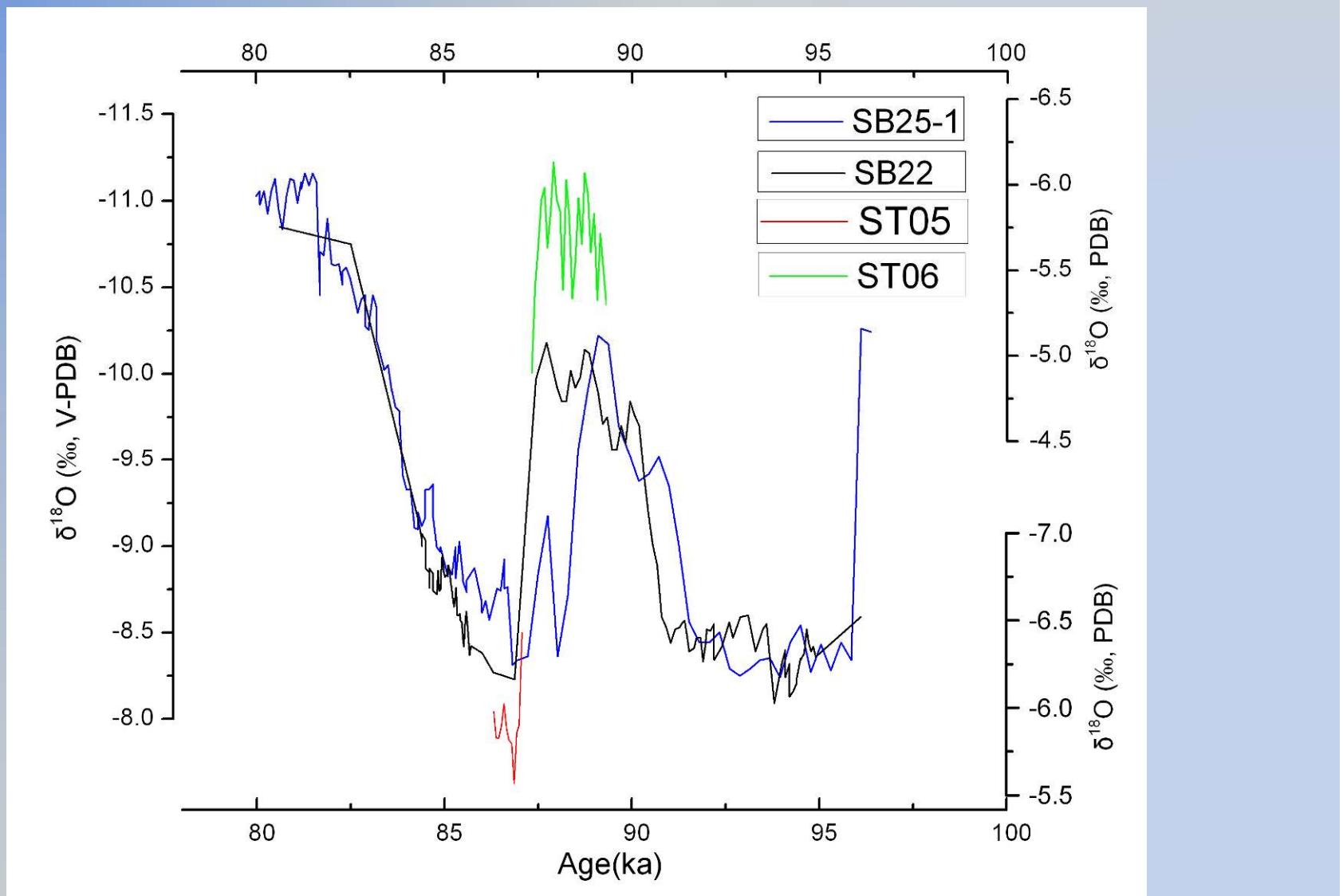


Song Terus and Tabuhan caves,
Java, Indonesia



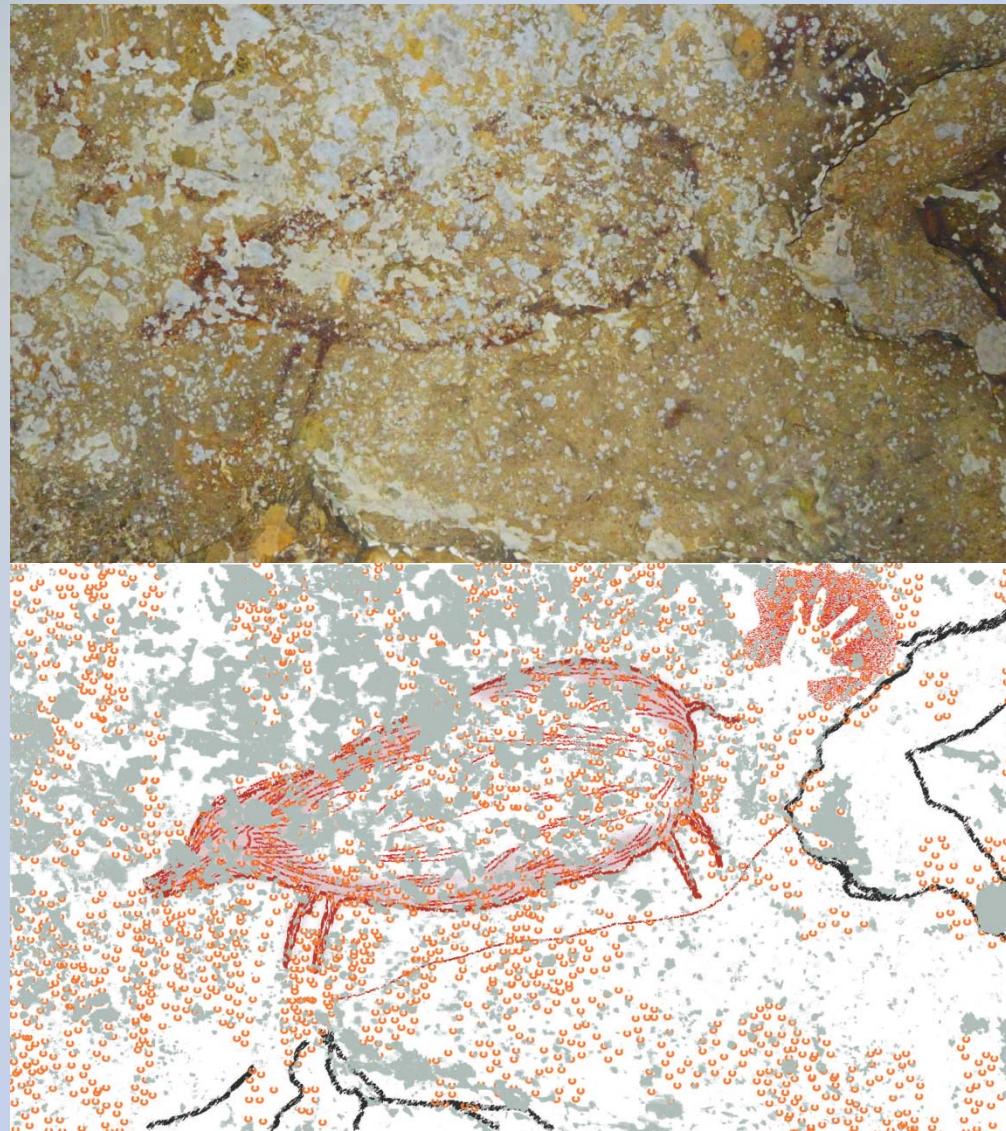
	238U	234U/238U	230Th/238U	230Th/232Th	Age[ka]
ST06-1	1.12	1.1398±0.0011	0.6557±0.0021	98.93	91.4±0.5
ST06-2	1.33	1.1406±0.0010	0.6050±0.0018	287.49	80.9±0.4
ST06-3	1.14	1.1407±0.0013	0.6349±0.0030	152.45	86.9±0.6
ST06-4	1.16	1.1412±0.0026	0.6224±0.0023	204.11	84.3±0.6
ST06-7	1.21	1.1428±0.0008	0.6291±0.0028	94.46	85.5±0.6
 ST05-1	 2.99	 1.1168±0.0012	 0.6319±0.0015	 2052.60	 89.4±0.4
ST05-2	3.50	1.1183±0.0010	0.6259±0.0025	10986.67	87.99±0.5
ST05-3	3.61	1.1127±0.0009	0.5962±0.0012	4845.07	82.43±0.3
ST05-4	3.64	1.1177±0.0015	0.9374±0.0034	2866.72	186.8±1.8
ST05-5	2.40	1.1116±0.0009	0.6882±0.0016	1539.94	103.1±0.4
ST05-a	3.48	1.103±0.020	0.529±0.016	345	80.9±3.8
ST05-6	3.53	1.1086±0.0009	0.6187±0.0044	4510.04	87.6±0.9
ST05-7	4.54	1.1066±0.0005	0.6005±0.0012	2596.66	84.4±0.2
ST05-8	3.58	1.1055±0.0007	0.6304±0.0022	4196.88	90.6±0.5
ST05-9	3.65	1.1030±0.0007	0.6203±0.0014	9352.38	88.7±0.3

(TU Hua., M2 Master, 2012, unpublished data)

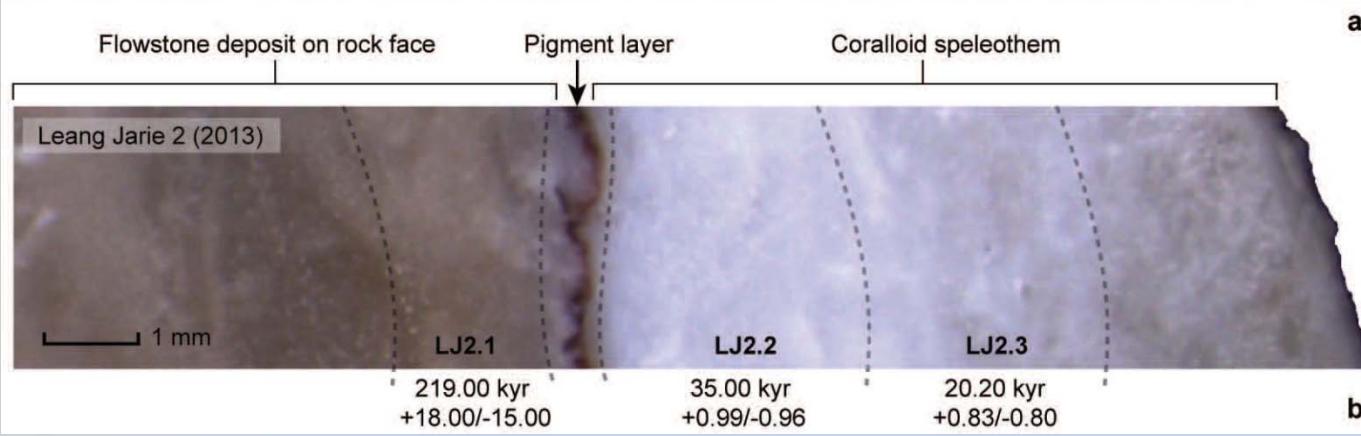
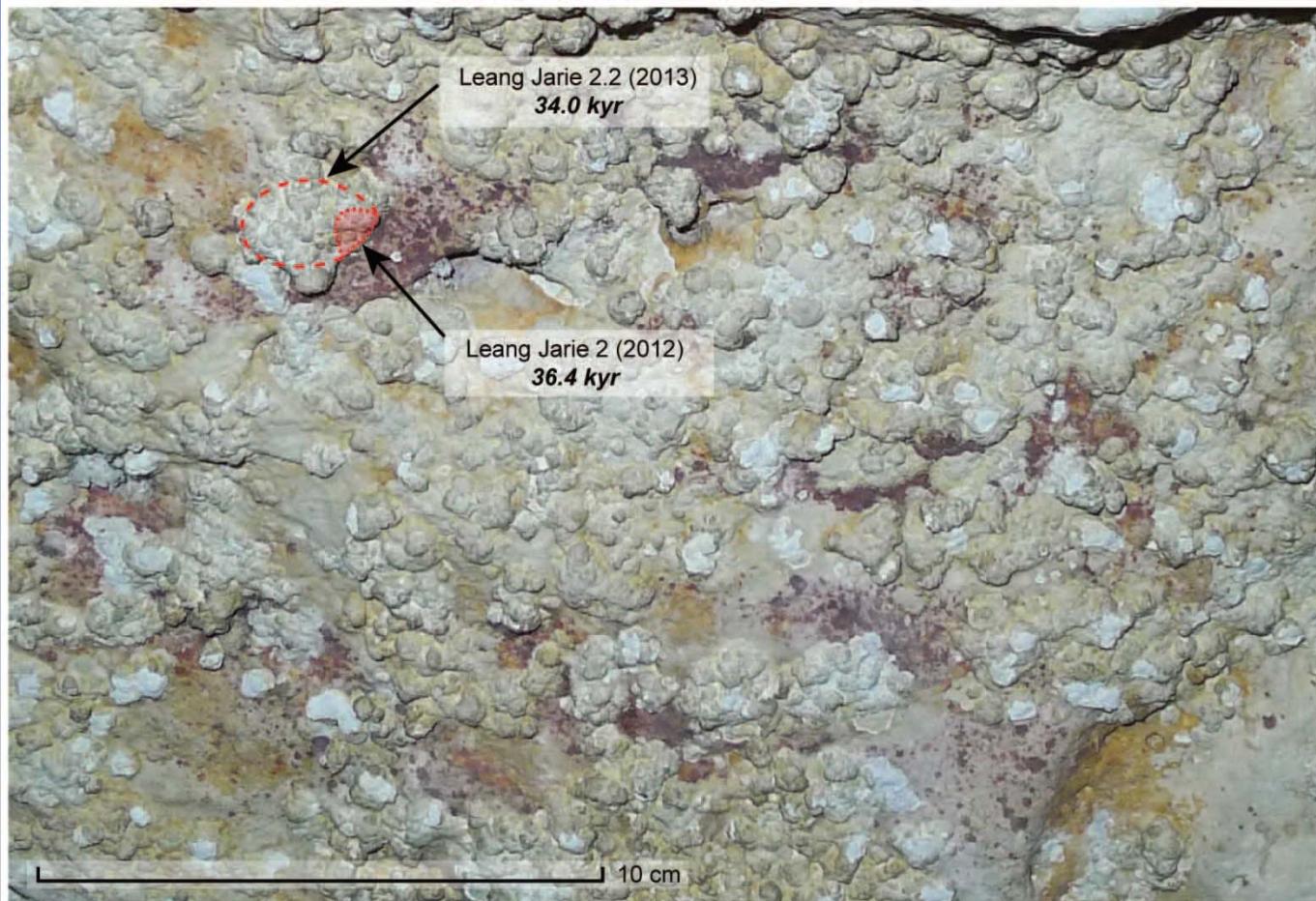


Comparison of isotopic records between Song Terus and Sanbao caves within around 95-80 ka. The left vertical axis corresponds to Sanbao curves, and the right ones are for ST06 (top) and ST05 (down).

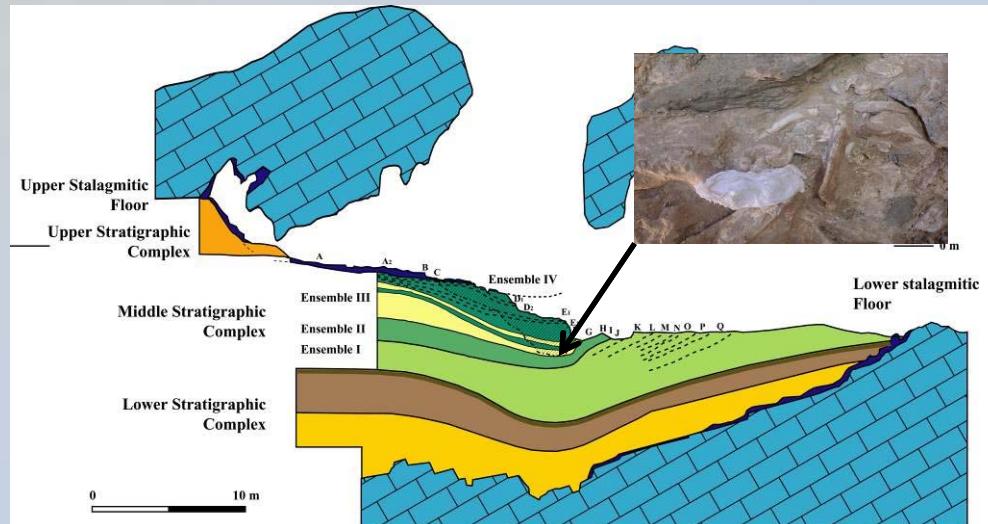
Dating valuable paintings in Sulawesi, Indonesia



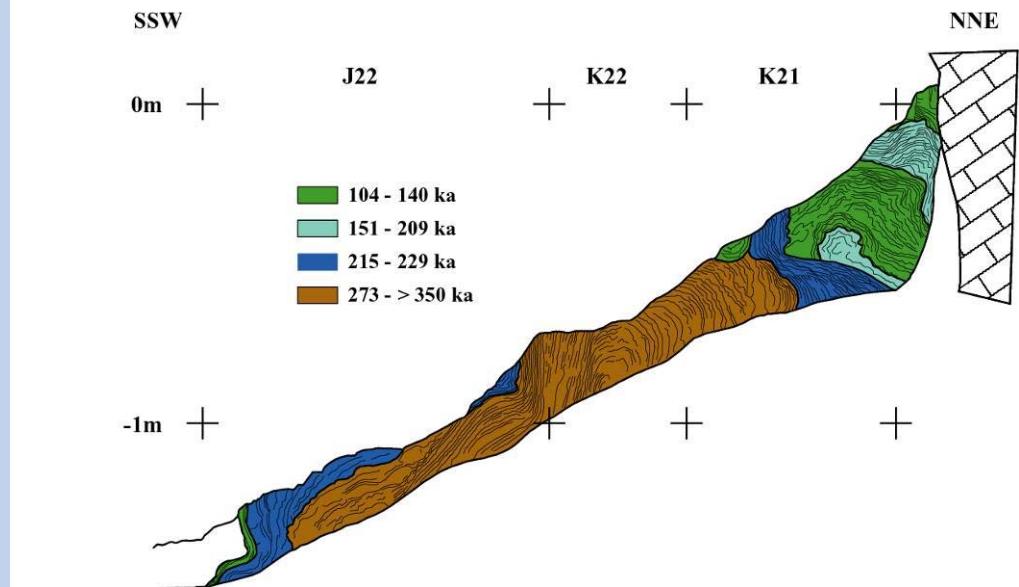
(Aubert et al., Nature, 2014)

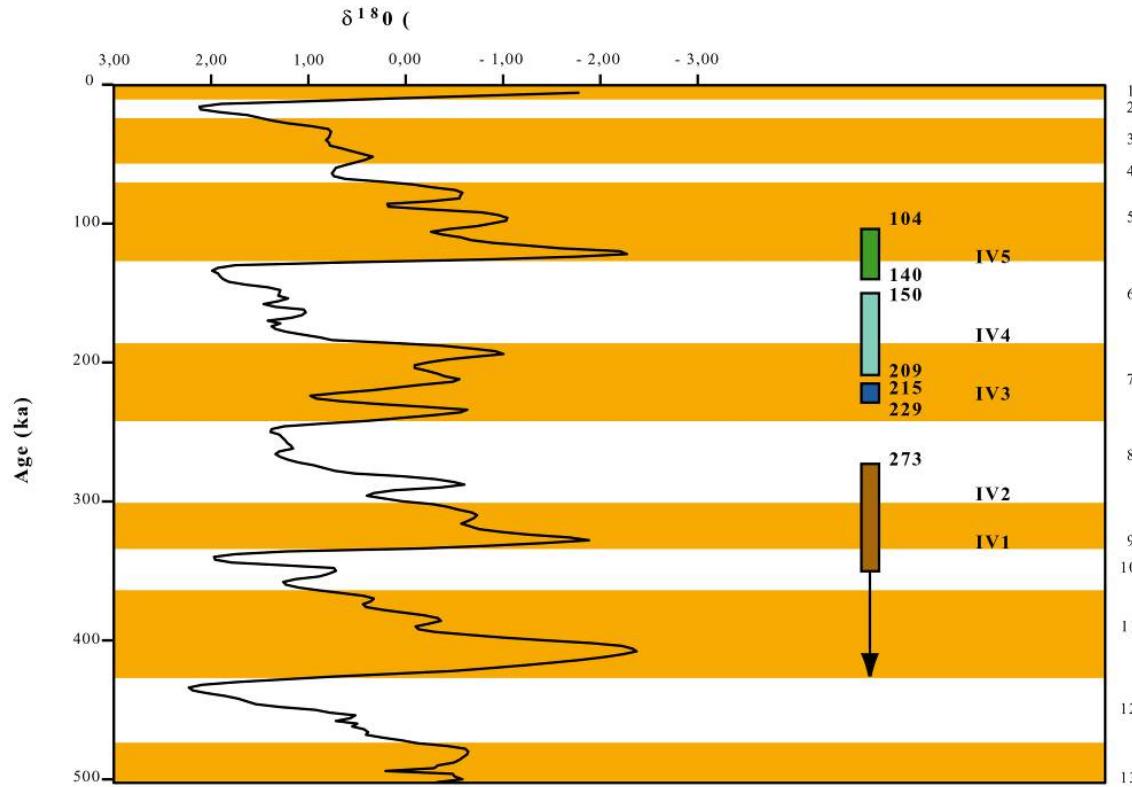


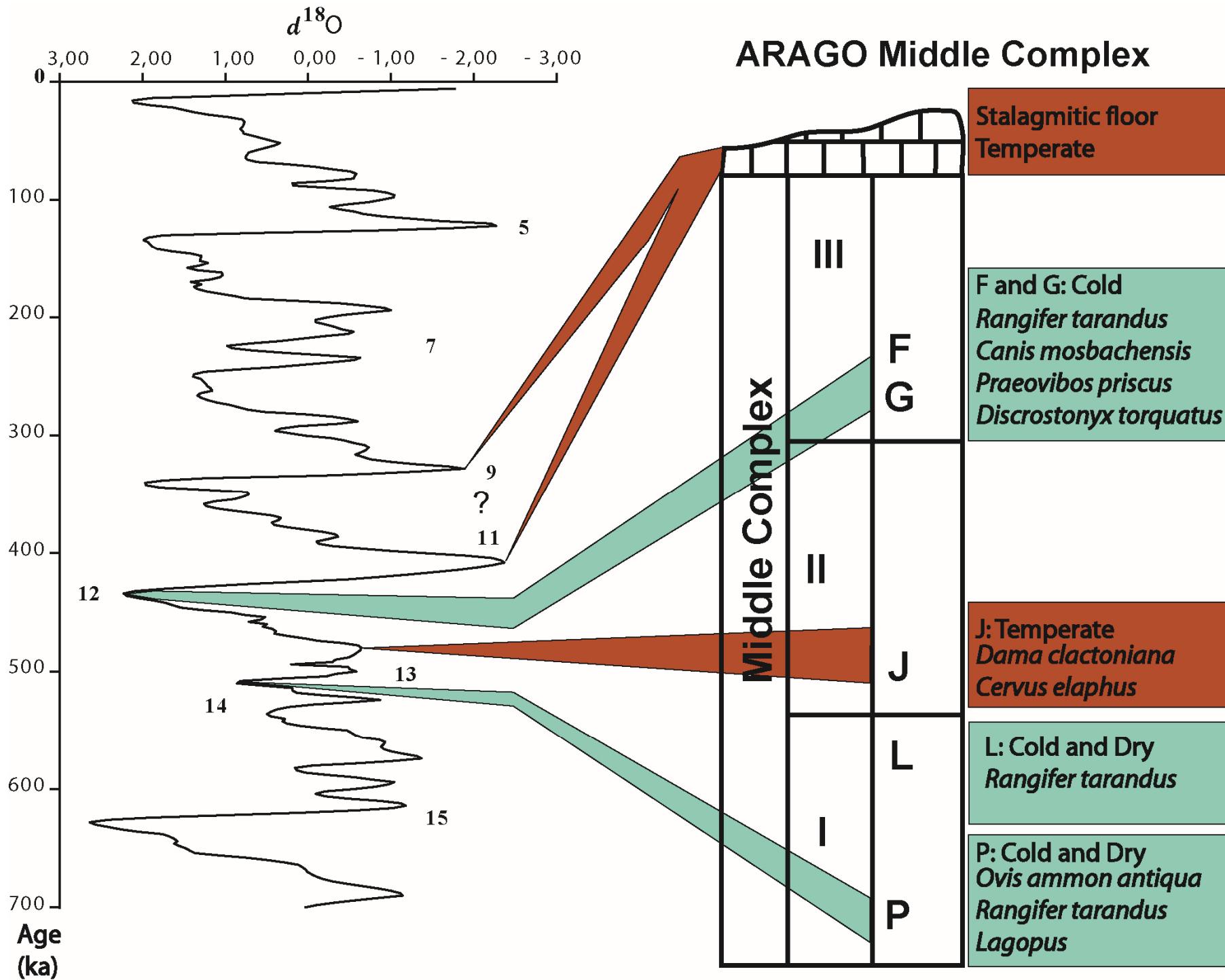
ARAGO CAVE, FRANCE



Uranium-Thorium - applications - Datation de la Caune de l'Arago







MARINE CARBONATES



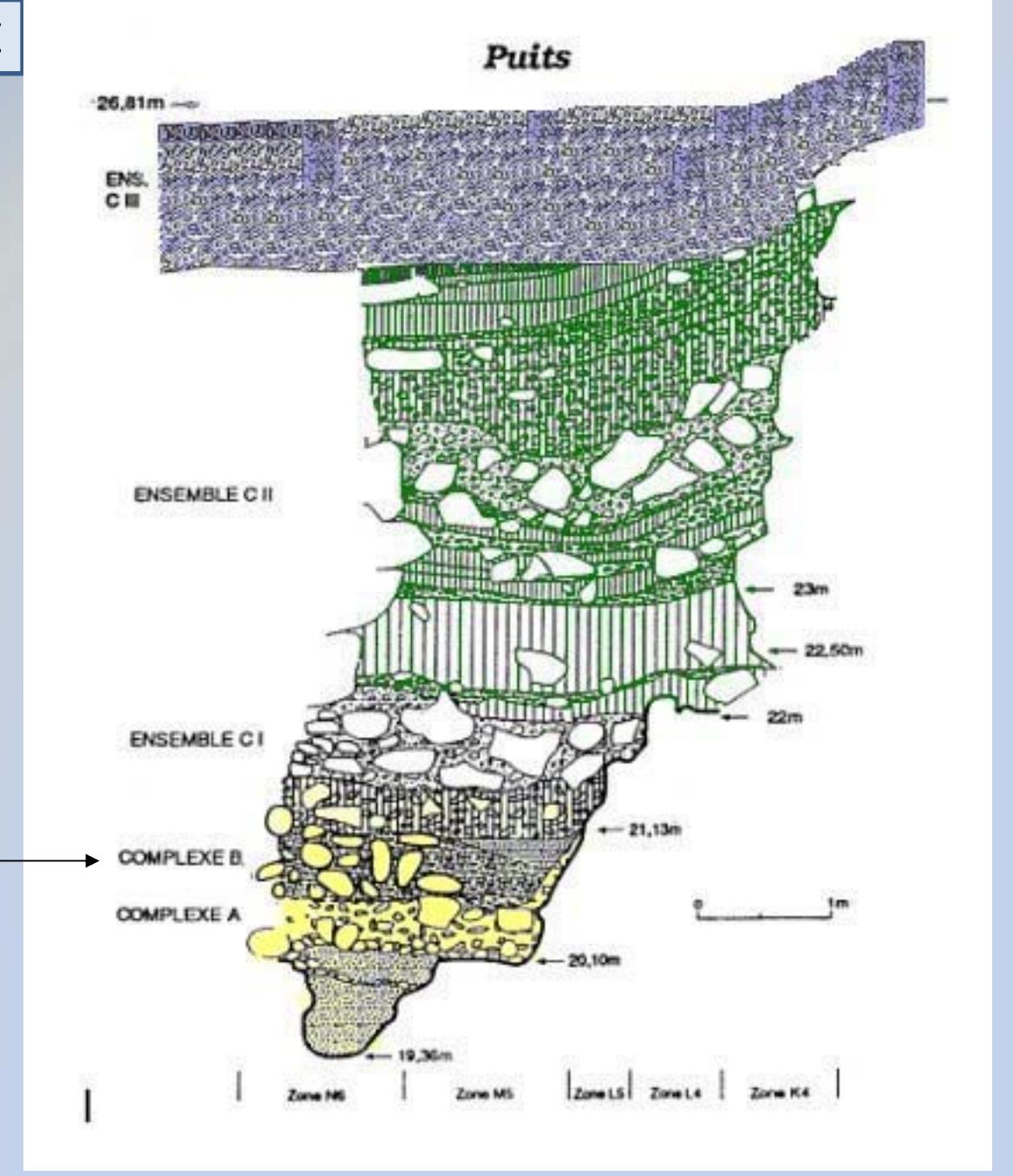
Shells and corals



LAZARET CAVE, NICE, FRANCE

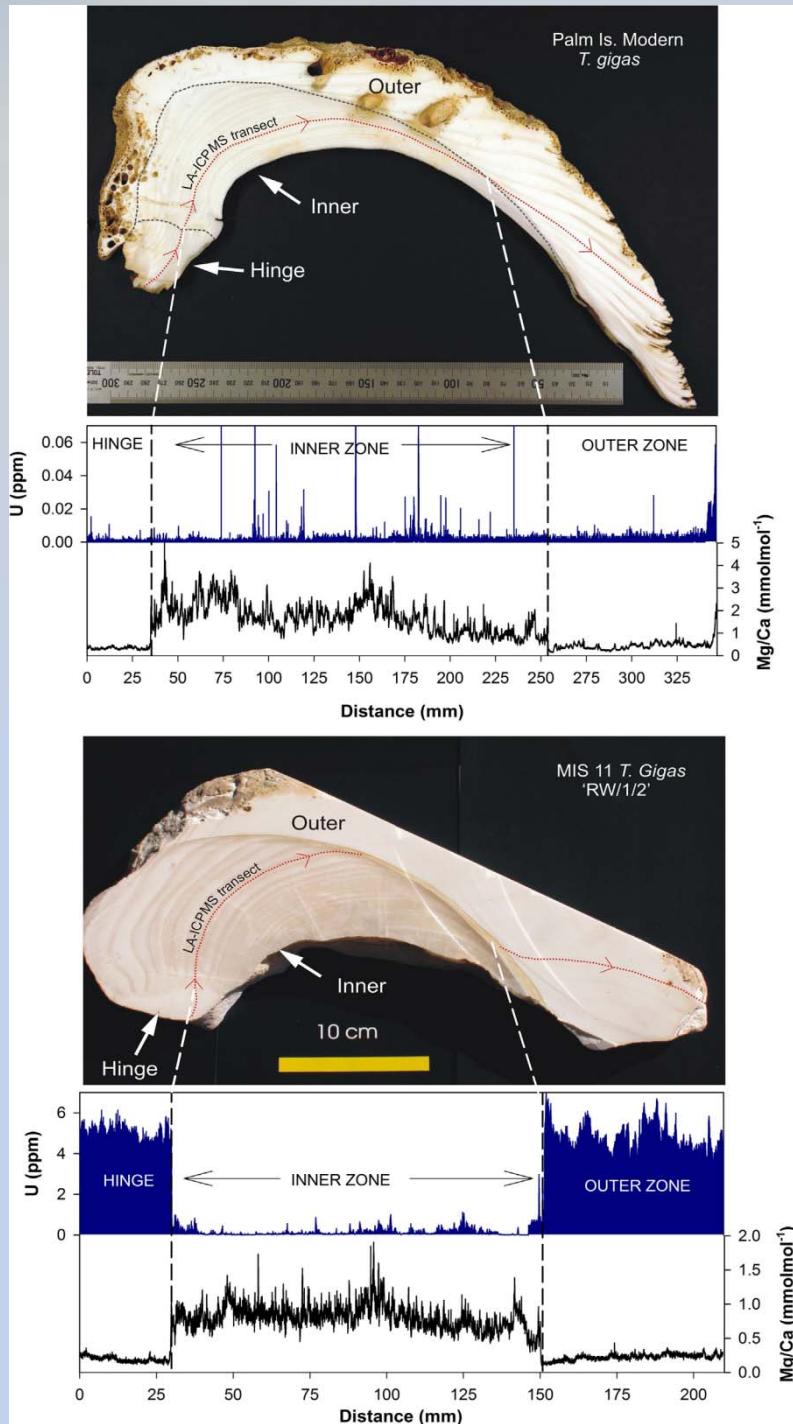


238 +22/-18 ka
Corals



Profiles of uranium concentrations in modern and MIS 11 *T. gigas* specimens, spanning the hinge, inner and outer zones (data acquired using LA-ICPMS).
(Ayling et al., GCA, 2017)

Zonal differences in U concentration are attributed to crystallographic differences between the growth zones, which are likely to affect the specific surface area of the aragonite crystal lattice and thus the availability of surface binding sites for uranium adsorption.



Cosmogenic isotopes

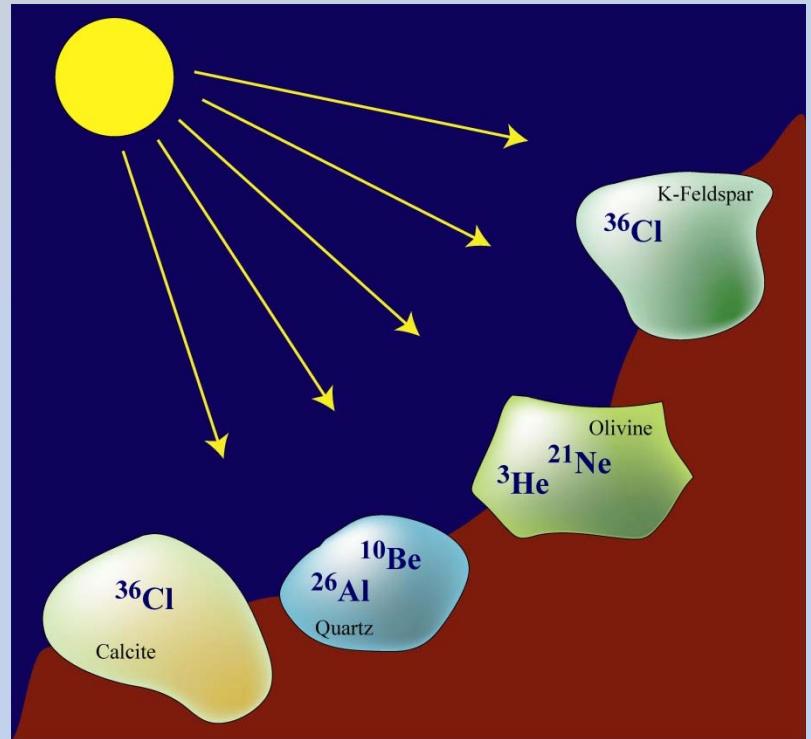
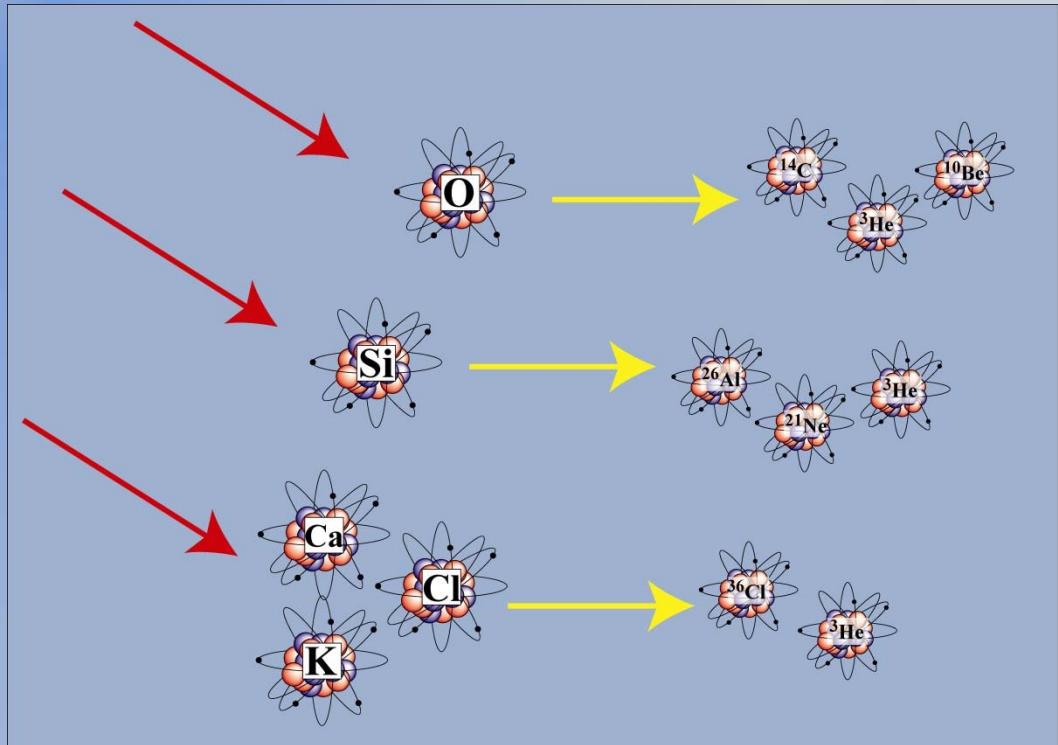
The dating of exposure by cosmogenic isotopes is a geochemical dating method which uses the production of rare isotopes by the cosmic rays, then their accumulation in the minerals crystal lattice to determine an exposure age.

In paleoseismology, to date a surface shifted by a fault.

In geomorphology, to calculate erosion rates :
by using a couple of isotope and their respective half-life.

In geomorphology, to obtain the age of an alluvial terrace, a moraine or any other formation.

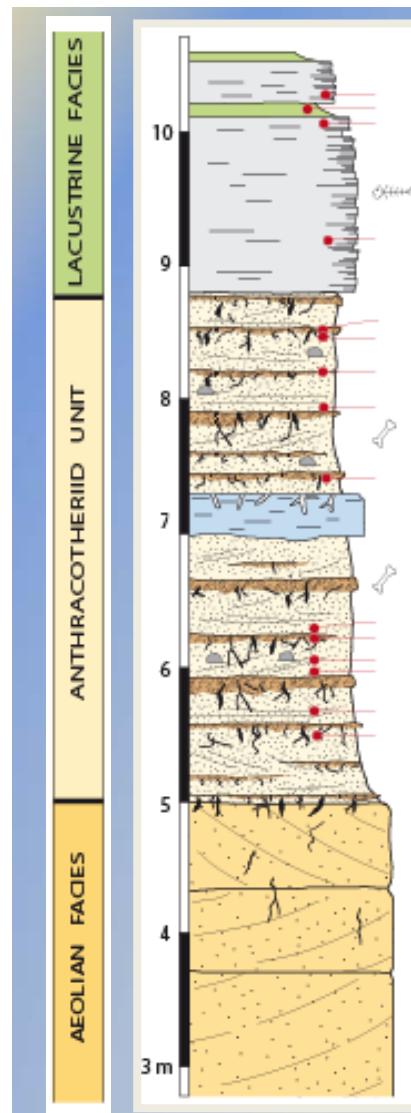
In paleoglaciology, to estimate a deglaciation :
the exposure starts when the ice does not cover any more the rock.



For instance, helium (^4He), Beryllium (^{10}Be), aluminium (^{26}Al), chlore (^{36}Cl), carbon (^{14}C), neon (^{21}Ne)

The isotopes production rates depend on:

- Altitude
- Latitude
- Depth (thickness of rock, water and/or snow to the top of the sample)
- Angle of incidence (angle enters the vertical and the sample)
- Masking related to topography (cliff, mountain...)



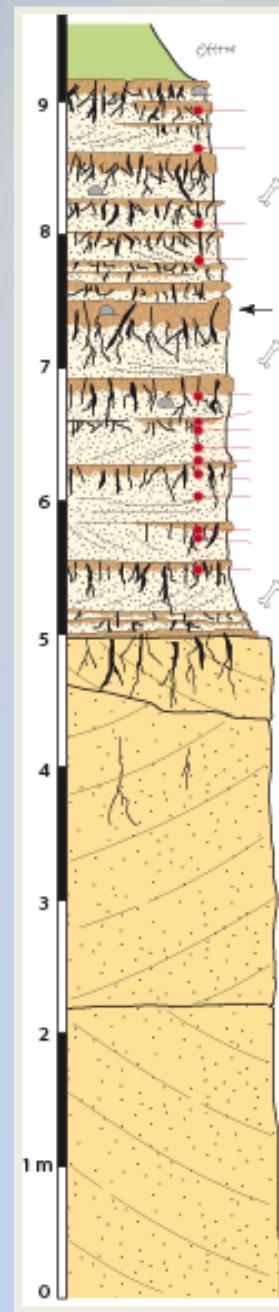
(Lebatard et al., PNAS, 2008)

6.25 ± 0.15 Ma

7.13 ± 0.14 Ma

7.46 ± 0.13 Ma

TM 266



S. tchadensis
Holotype, cranium

7.10 ± 0.14 Ma

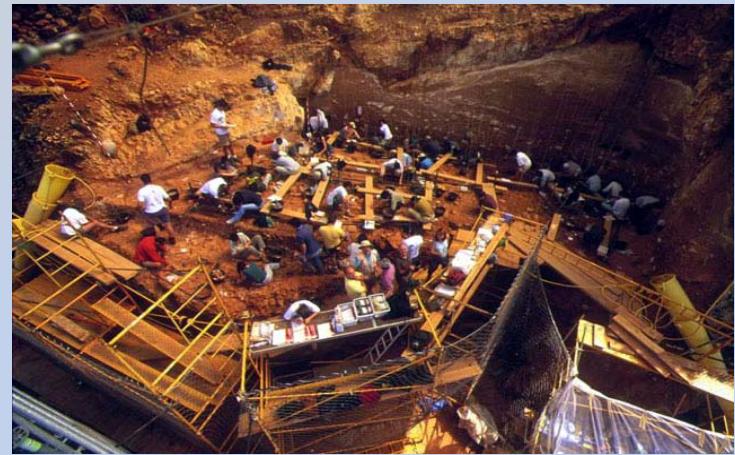
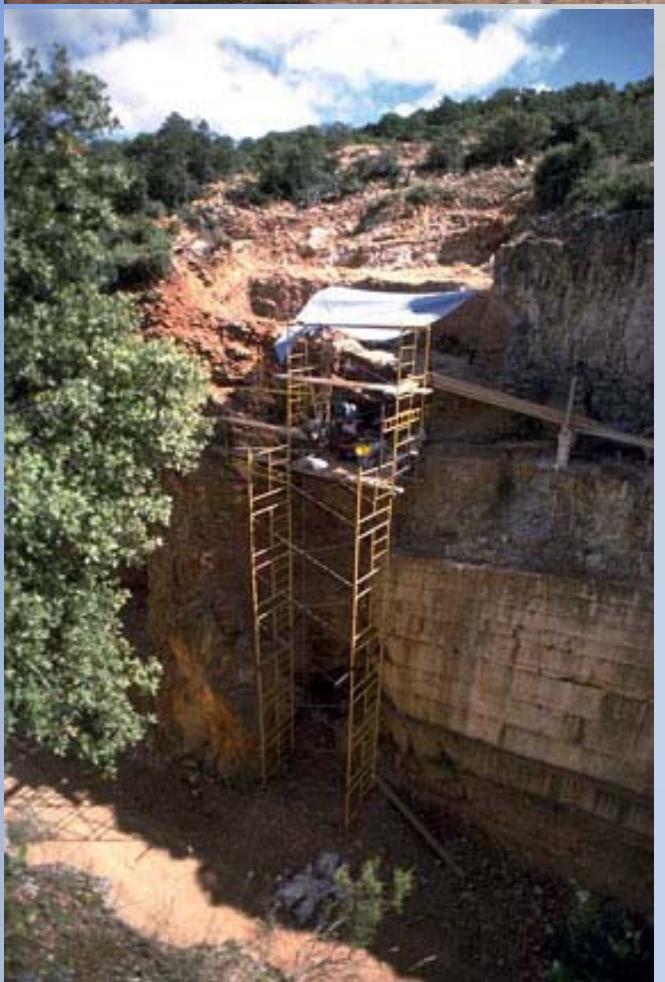
7.34 ± 0.12 Ma



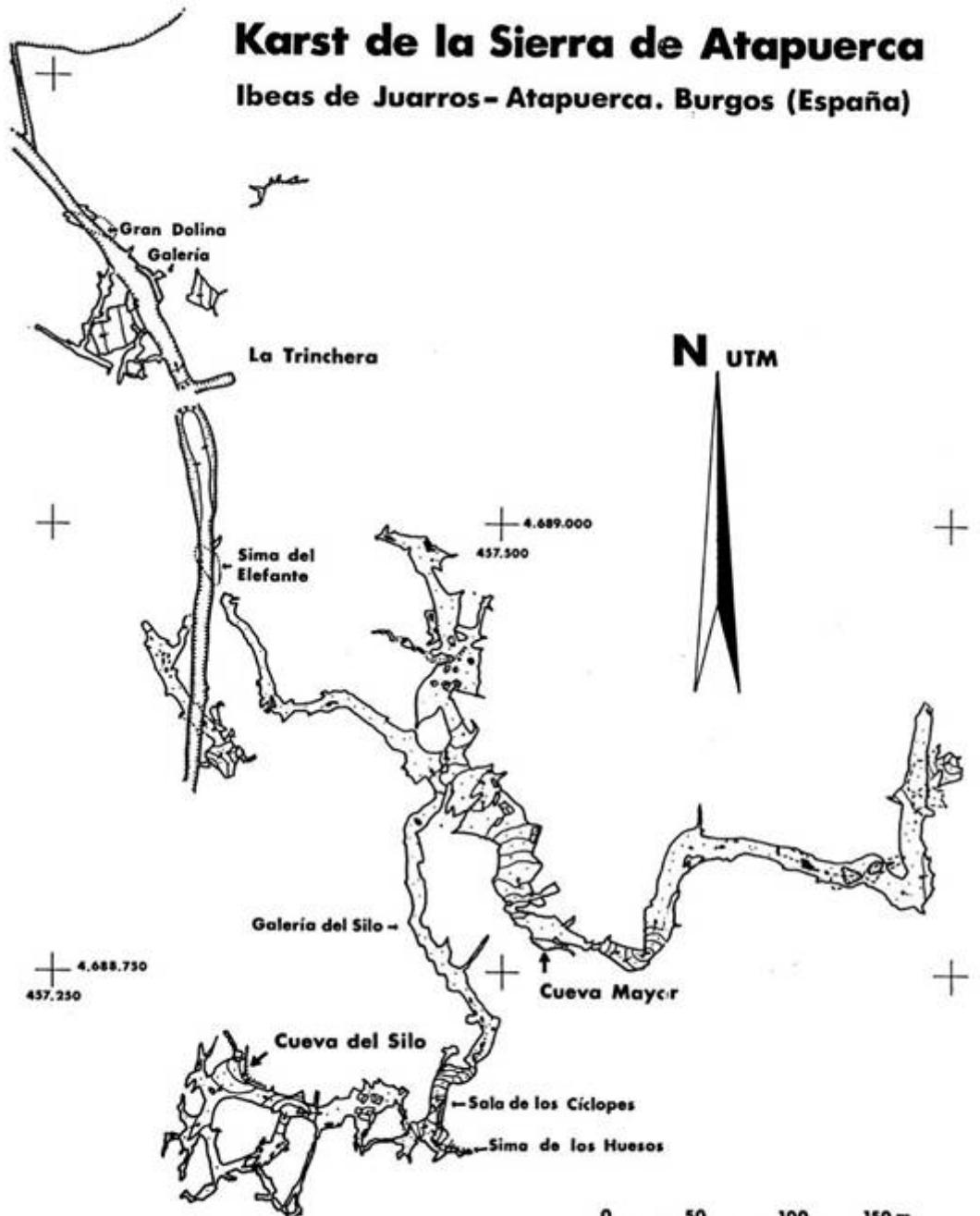
Sahelanthropus tchadensis
about 7 Ma

Dating Toumai, Tchad

^{10}Be : $T_{1/2} = 1,4 \cdot 10^6$ ans
Authigenic $^{10}\text{Be}/^{9}\text{Be}$



Sierra de Atapuerca, Espagne



G. E. Edelweiss



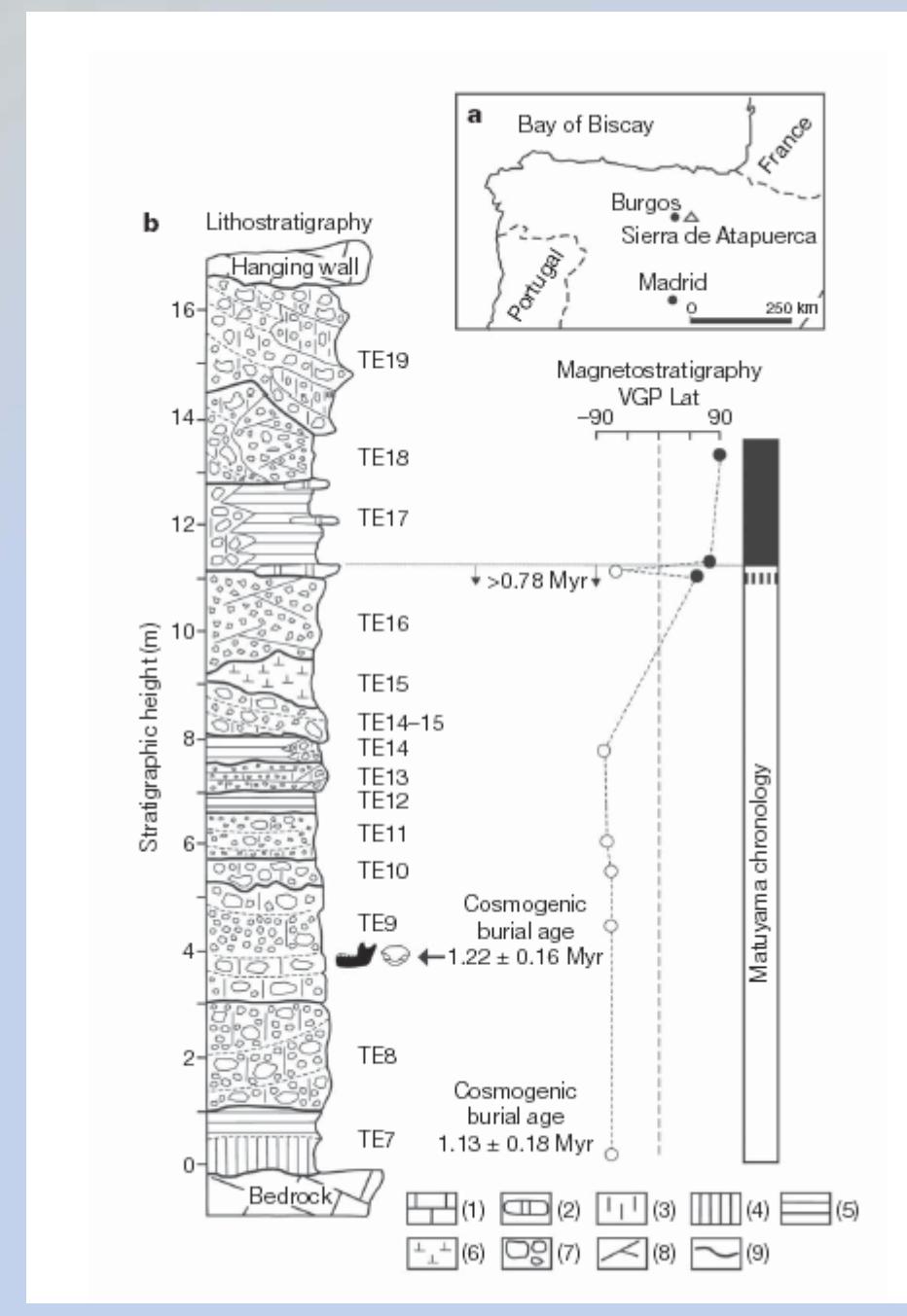
Lám. XVIII.—Entrada a la cueva de Atapuerca, en Ibeas de Juarros.



Sima del Elefante



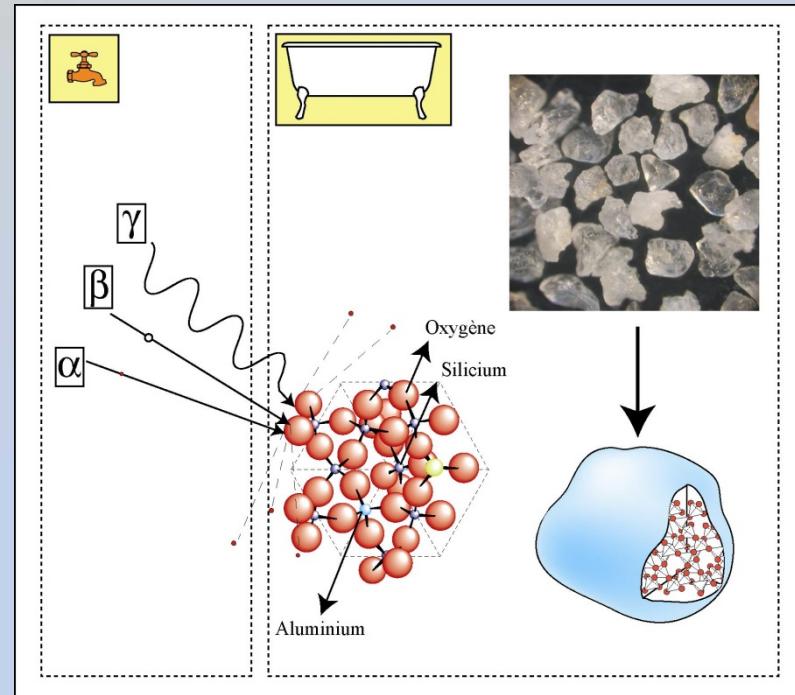
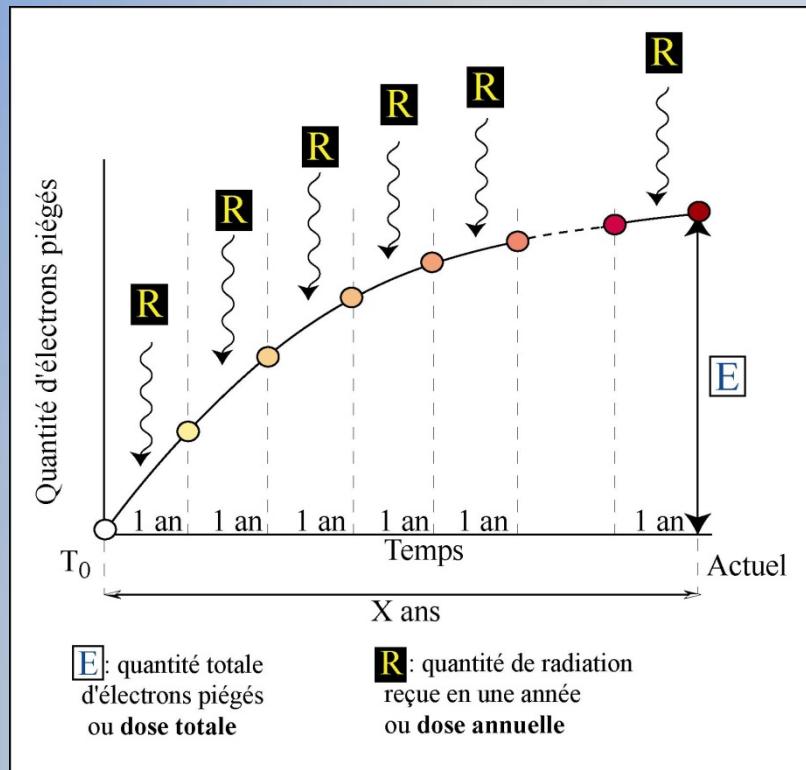
(Carbonell et al., Nature, 2008)



Palaeodosimetric Methods

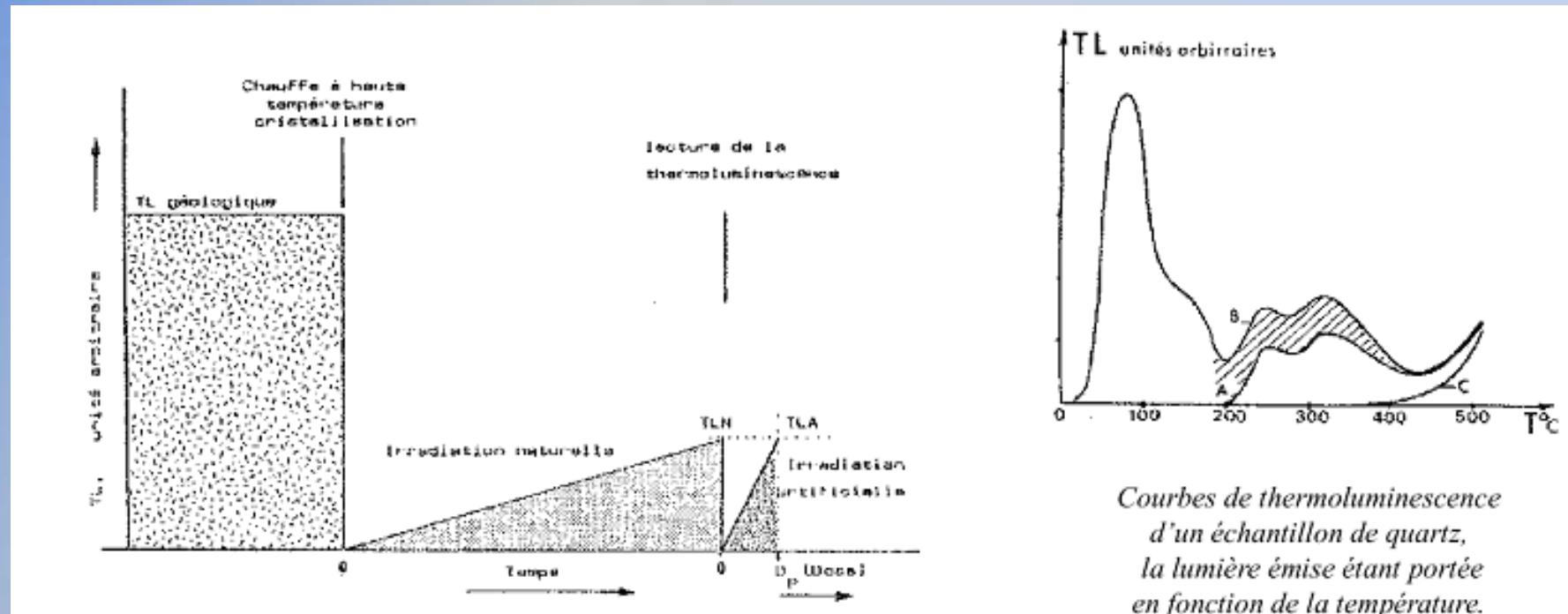
Résonance de Spin Electronique

Méthodes de la Luminescence



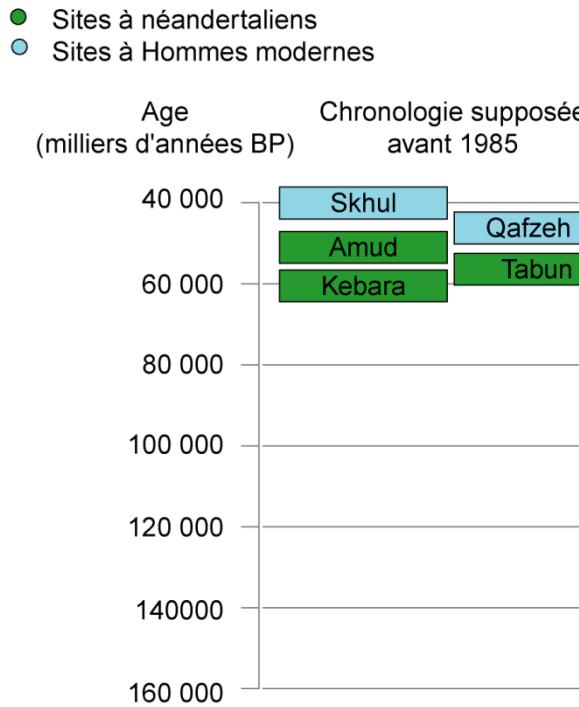
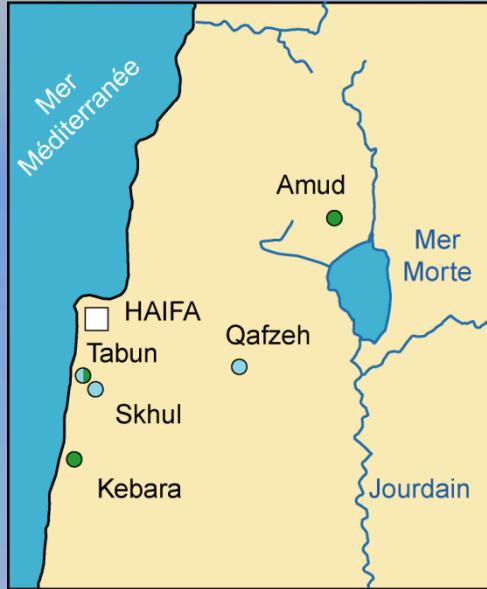
$$X \text{ ans} = \text{Age} = \frac{E}{R}$$

Thermoluminescence



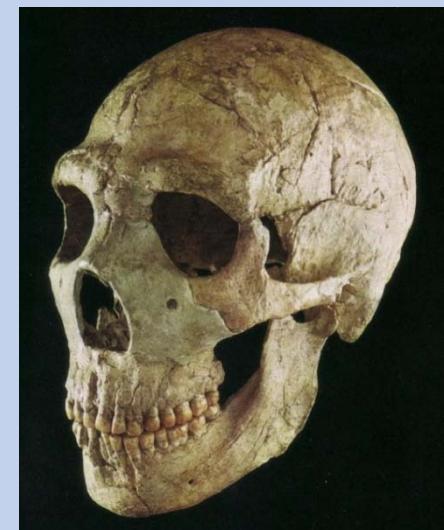
Last dated event = last heating
or light exposure

Main dating samples : burnt
flint, quartz, feldspar



Qafzeh 6

**Application -
Les sites à
néandertaliens et
hommes
anatomiquement
modernes du Levant**



AMUD

Valladas et al., Nature, 1987

OSL

Optically Stimulated Luminescence

Last dated event = last light
exposure

Main datable samples :
quartz, feldspars

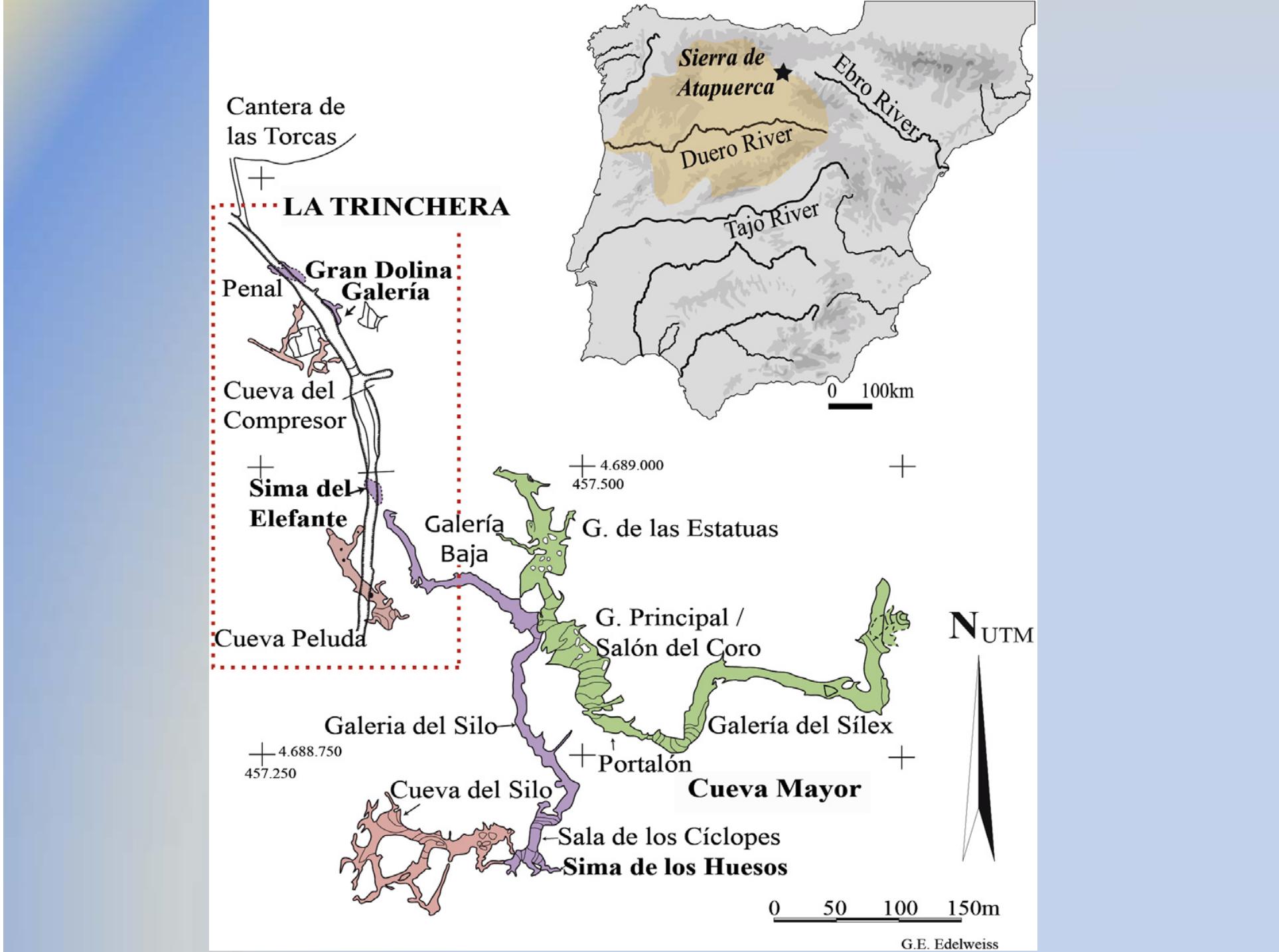
Applicability

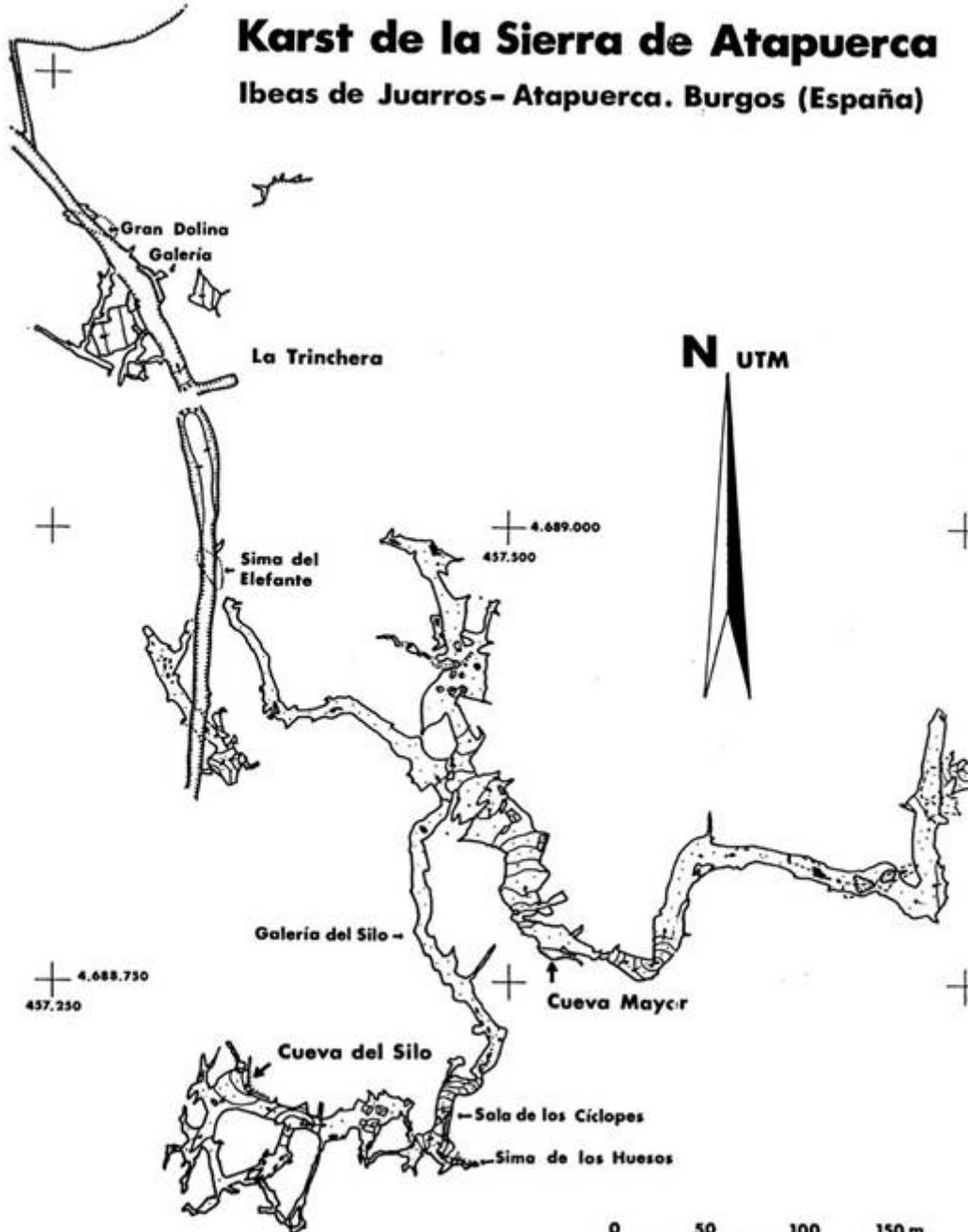
OSL: Holocene, Upper Pleistocene

IRSL: Holocene Upper and Middle Pleistocene

TT OSL: Lower and Middle Pleistocene

PostIr-Ir: Lower and Middle Pleistocene





G. E. Edelweiss



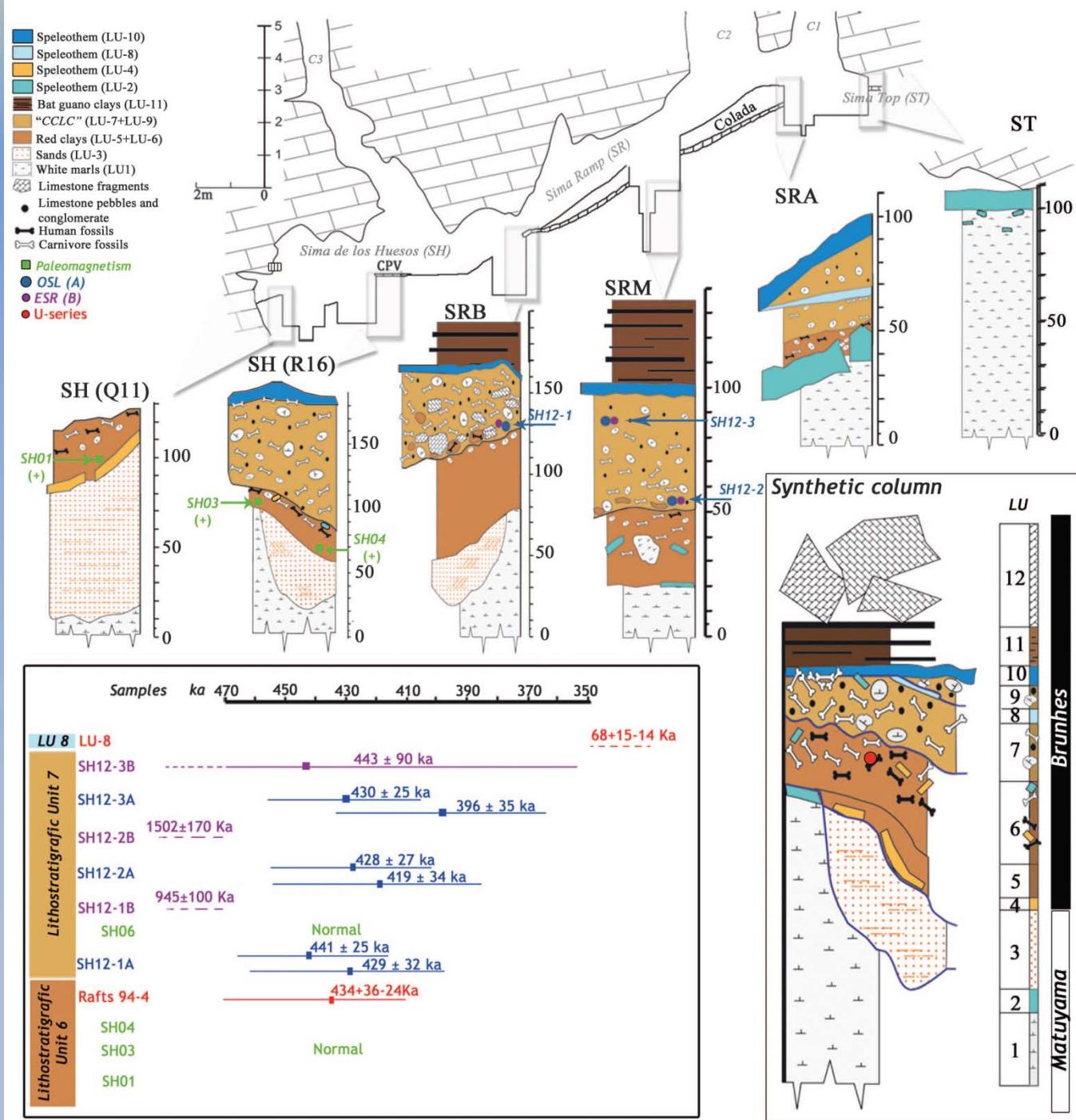
Lám. XVIII.—Entrada a la cueva de Atapuerca, en Ibeas de Juarros.





3 cm

Sima de los Huesos, Atapuerca



Measurement of Palaeodose DE

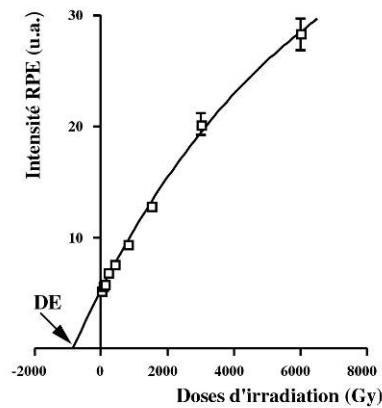
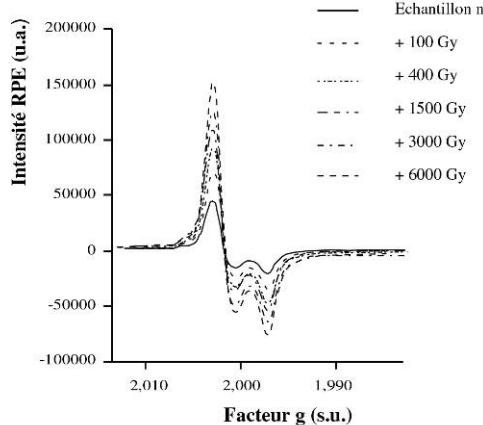


Figure 4 : Croissance du signal à l'irradiation et détermination de la DE par la méthode de l'addition en utilisant une extrapolation exponentielle.

ESR method

$$\text{AGE} = \text{DE} / \text{da}$$

Measurement of the annual dose, da

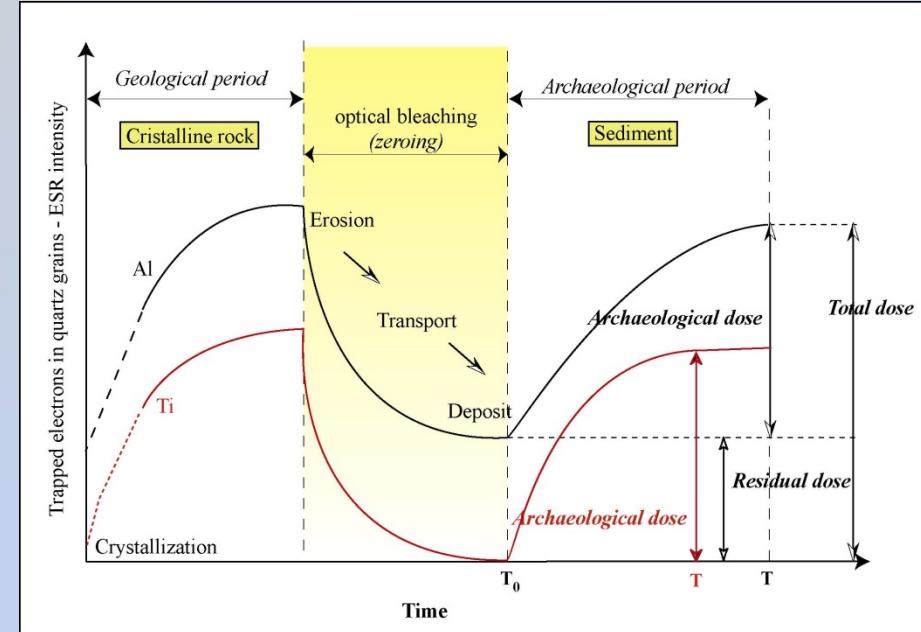
U, Th + daughters, K



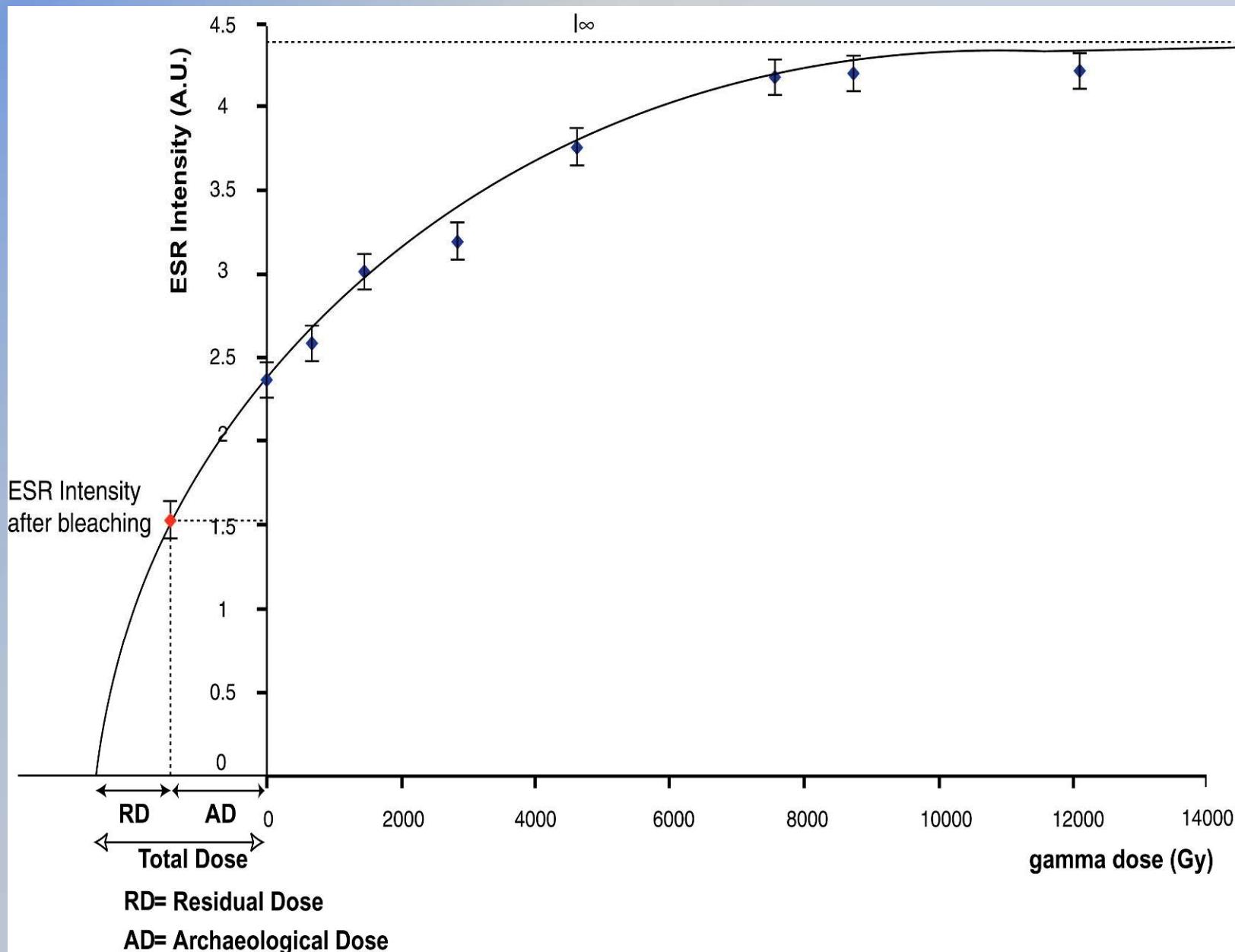
ESR Dating - Quartz



Frequent material

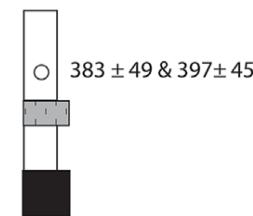
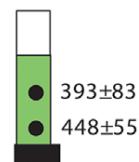


... but no complete bleaching

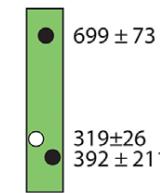


ENGLAND***Post MIS12 sites***

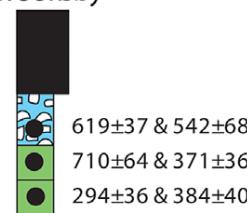
Beeches Pit

East Farm
Barnham

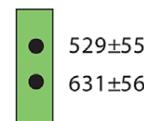
Purfleet

***Pre MIS12 sites***

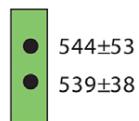
Brookby



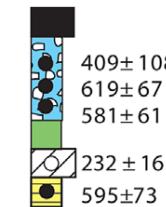
Maidcross Hill



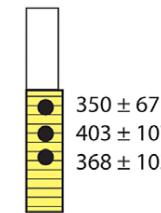
Warren Hill



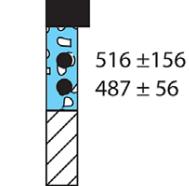
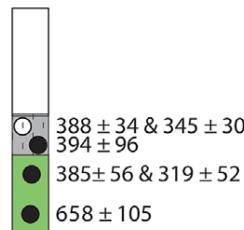
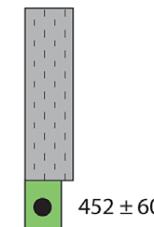
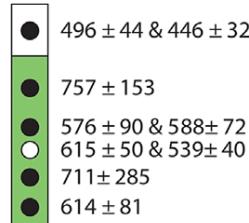
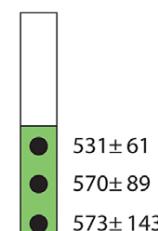
Pakefield



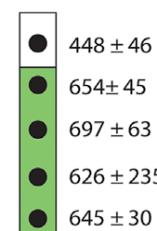
Valdøe



West Runton

**FRANCE*****Post MIS12 sites***Saint-Pierre-
lès-ElbeufLa-Celle-
sur-Seine***Pre MIS12 sites***Abbeville
Carrière CarpentierAmiens
Rue du Manège

La Noira



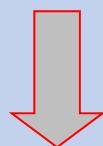
- ESR/U-series age on teeth
- ESR age on sediment
- Fluvial sediments
- ▨ Shallow marine sediments
- ▢ Fluvio-glacial sediments
- Cover deposits
- ▨ Tufa
- MIS12 Till
- ▨ Cromer Forest Bed (CfB)

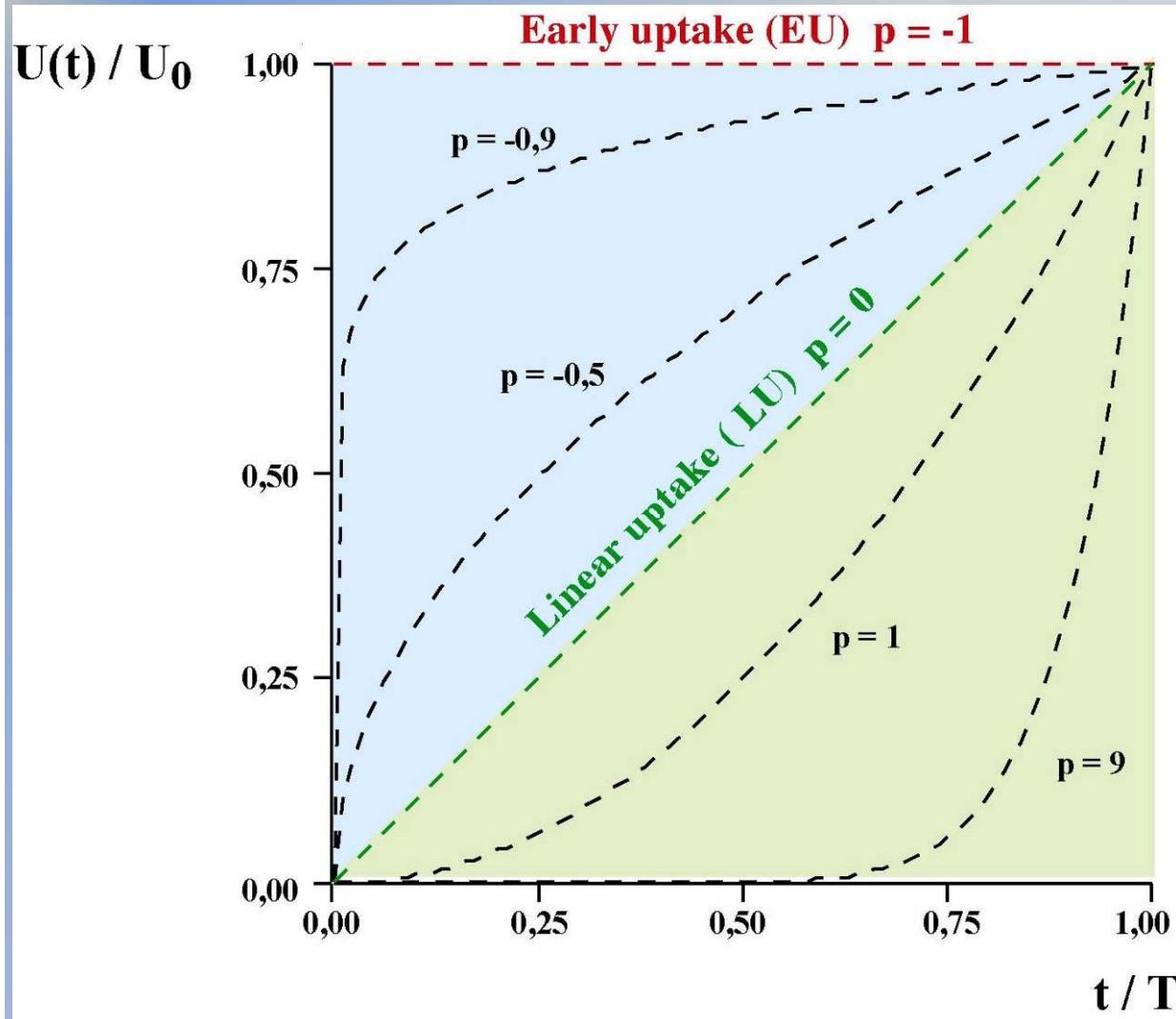
Combined ESR/U-Th Dating - Dental enamel

Direct Dating of human bearing occupation layers



but post-mortem uranium uptake
should be taken into account in the
annual dose calculation





$$U(t) = U_0(t/T)^{p+1}$$

ESR Ages < U-Th Ages
 \Rightarrow U leaching, no age calculation

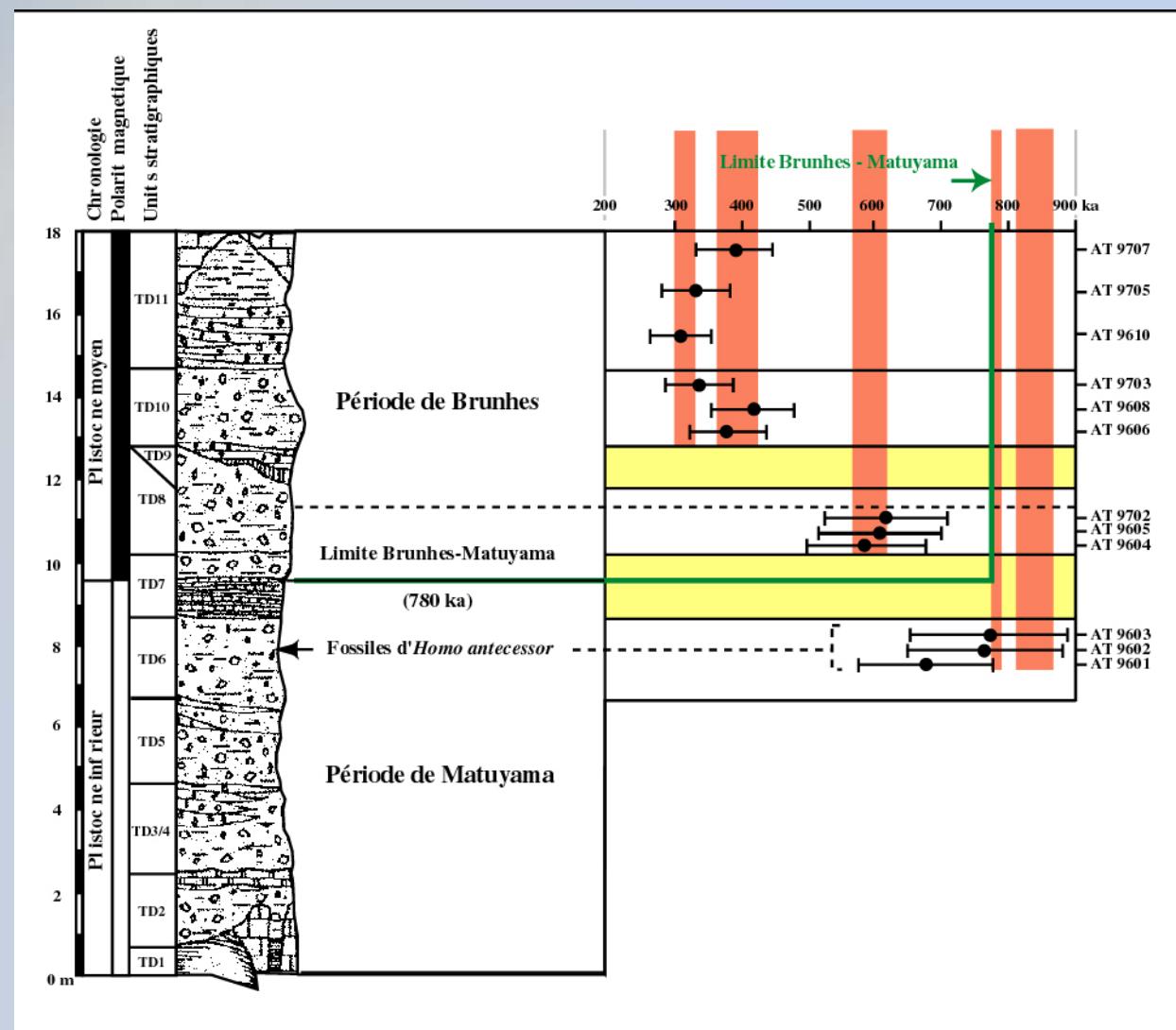
$-1 < p < 0$ area where ages are «acceptables»

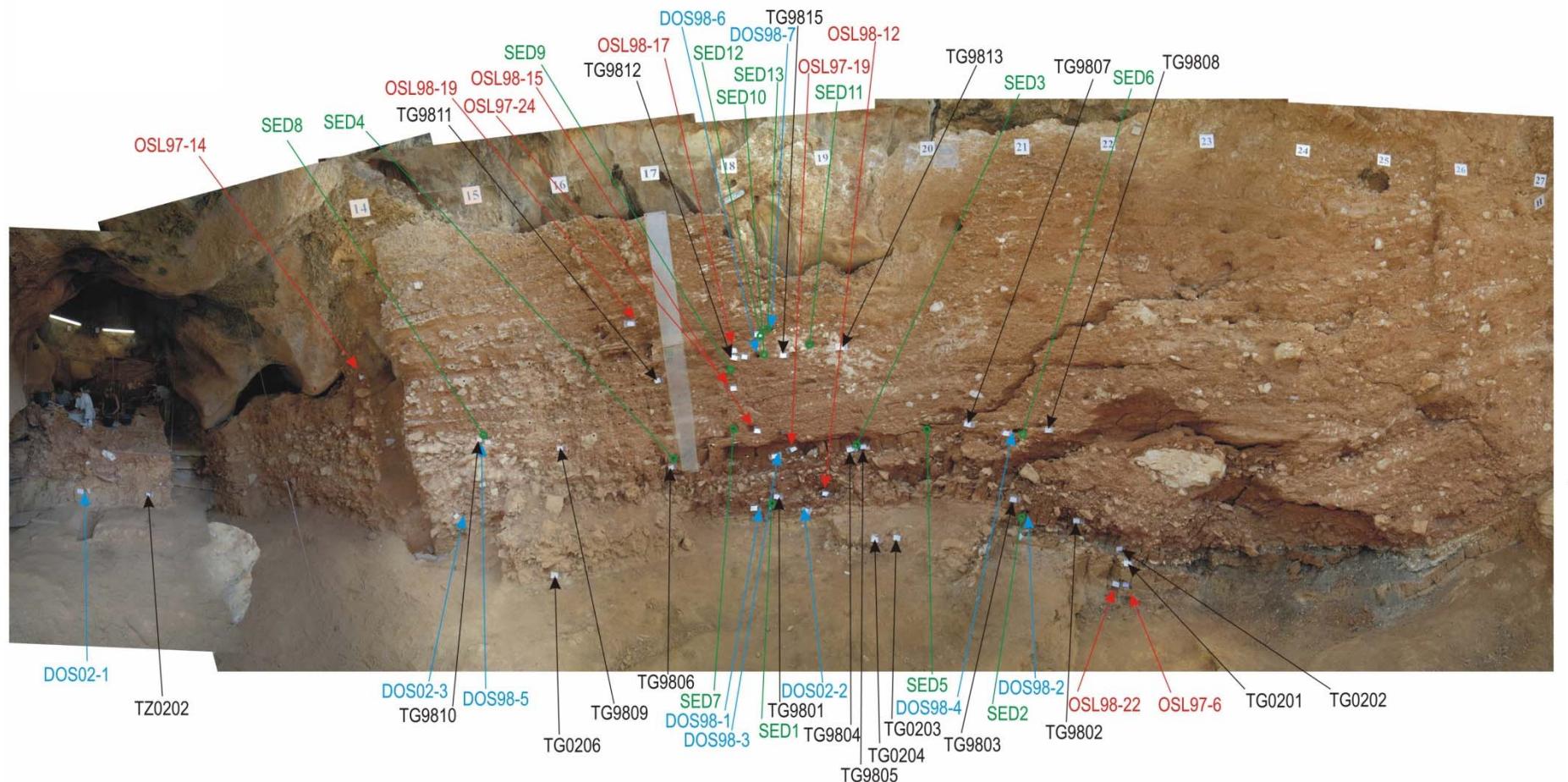
$p > 0$ recent uranium uptake

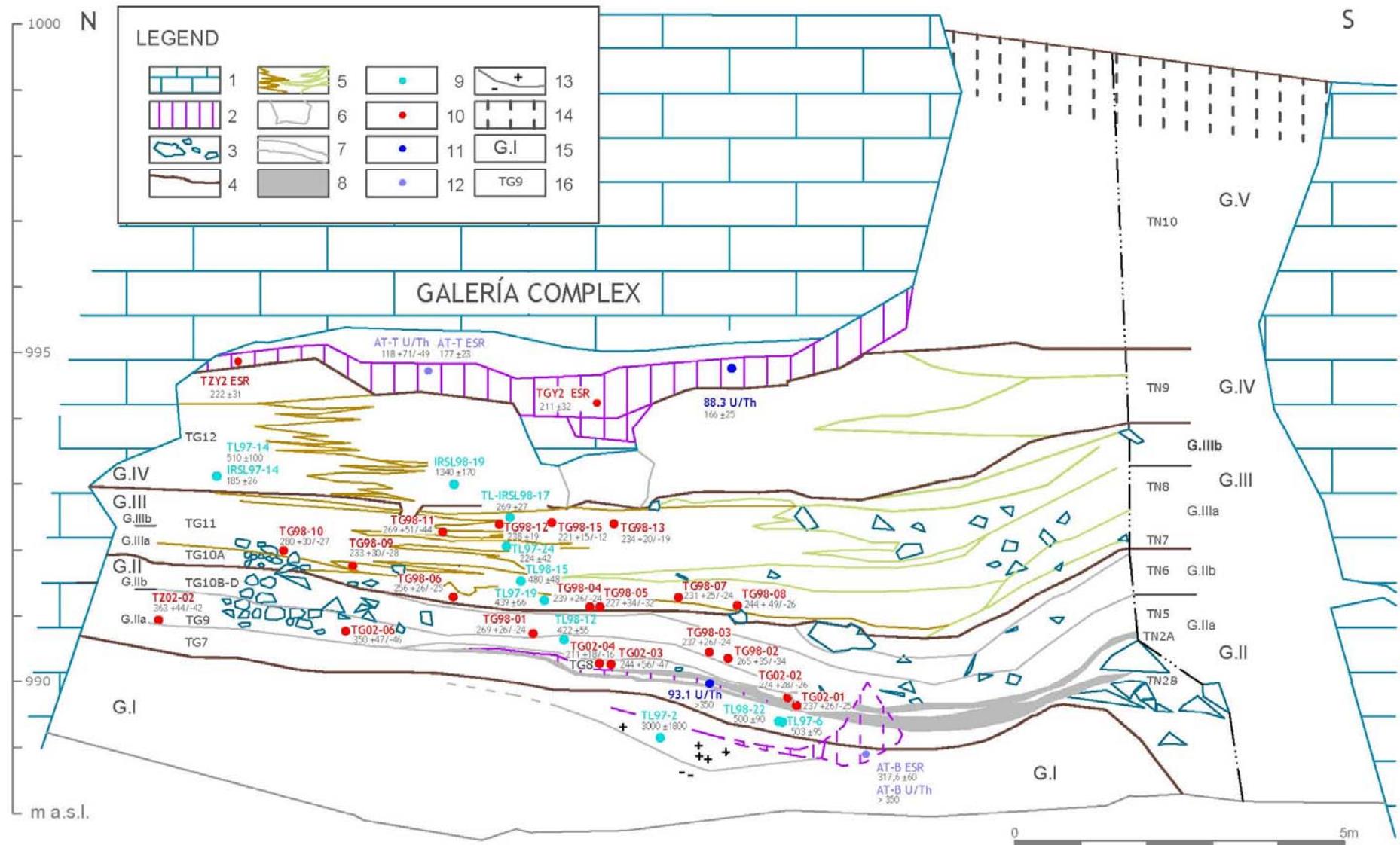
Grün et al., Nuclear Tracks, 1988

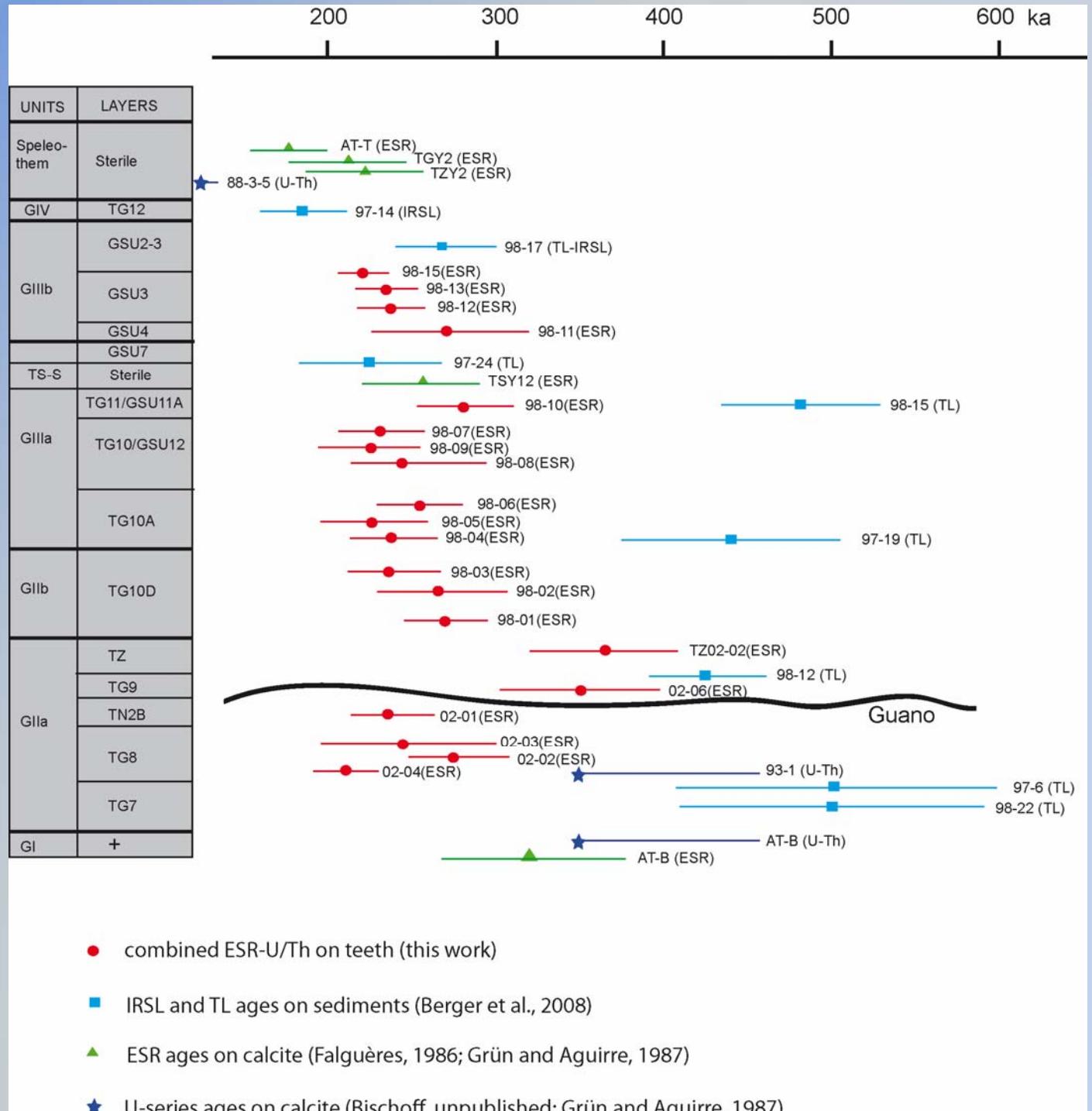


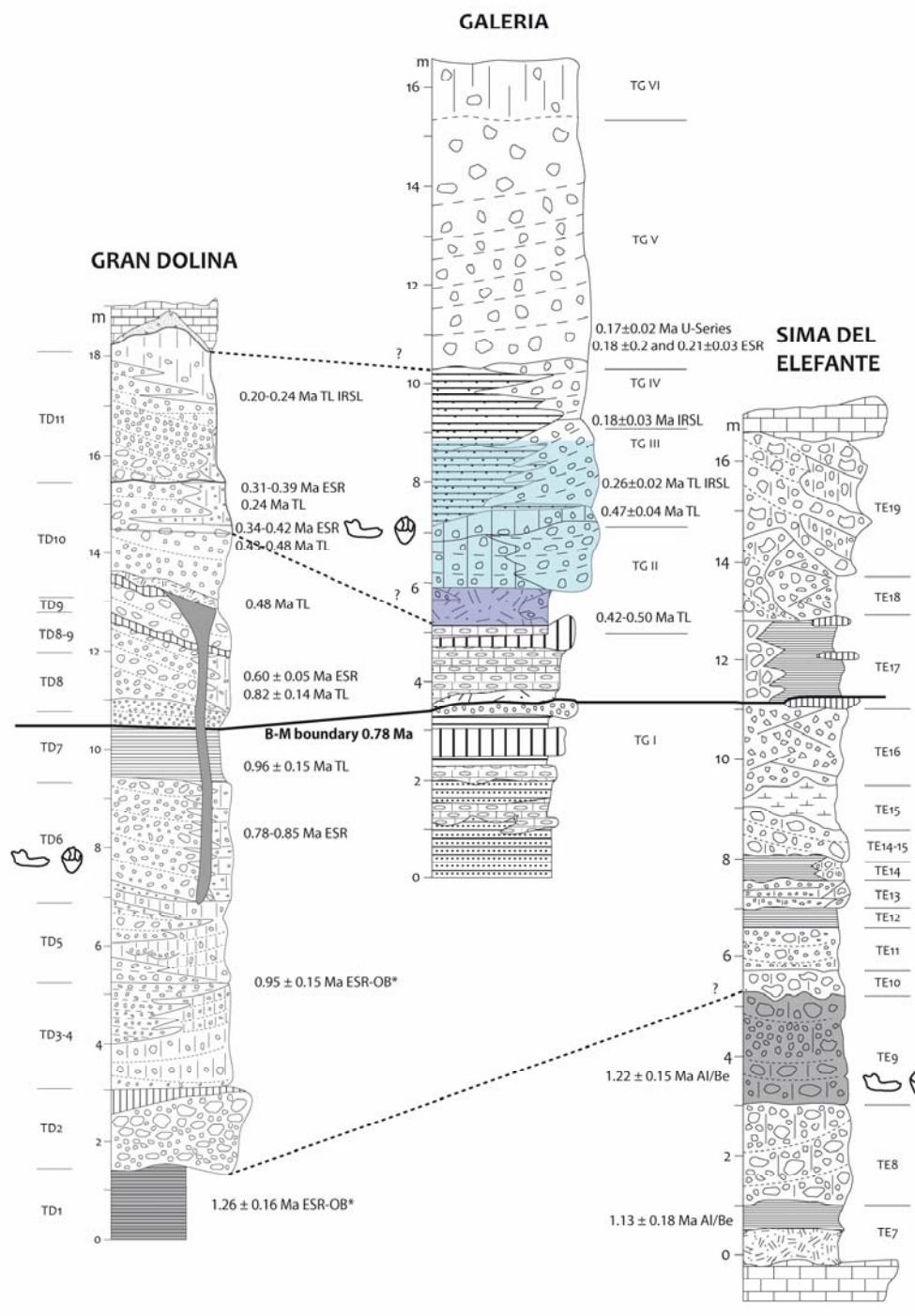
Application - Atapuerca Gran Dolina







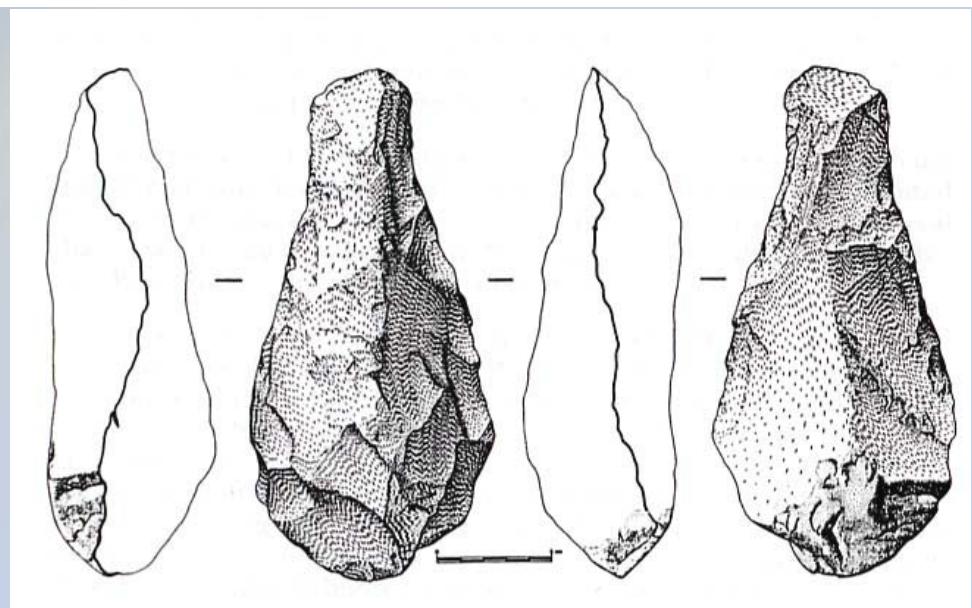




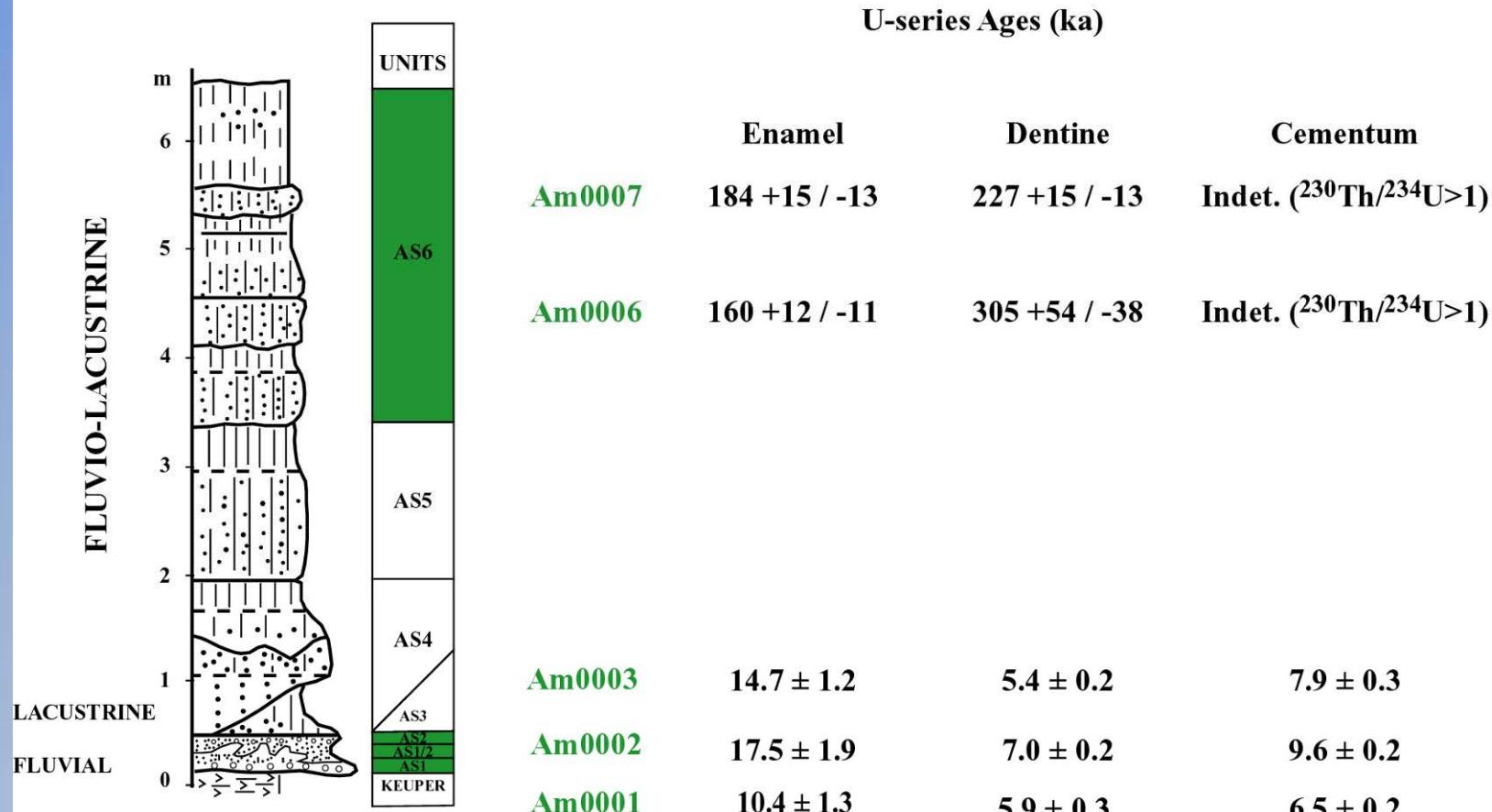


AMBRONA, SPAIN

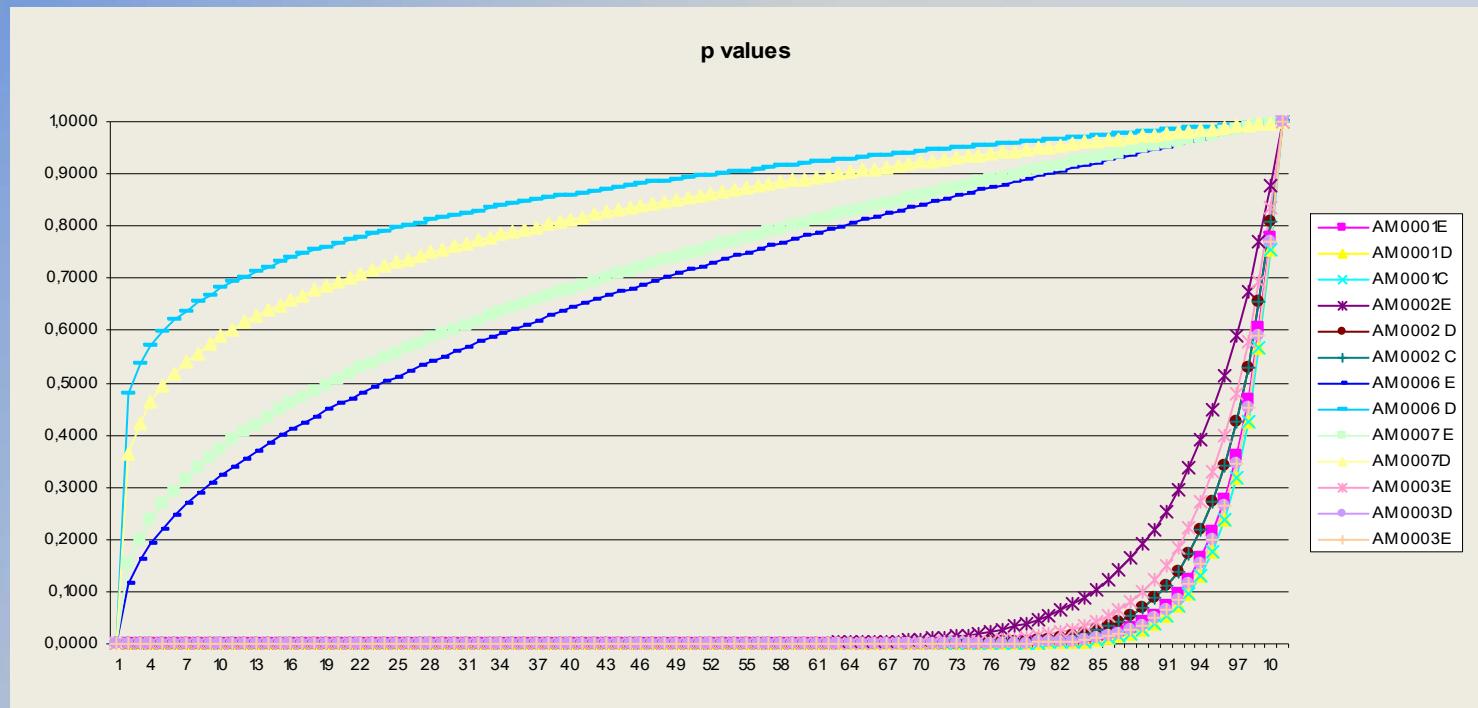
Elephas antiquus
89% of the remains
discovered in 1995
Cervidae
Bovidae
Equidae



AMBRONA



Lithostratigraphic units of the Lower complex of Ambrona (after Perez-Gonzalez et al., 1999)

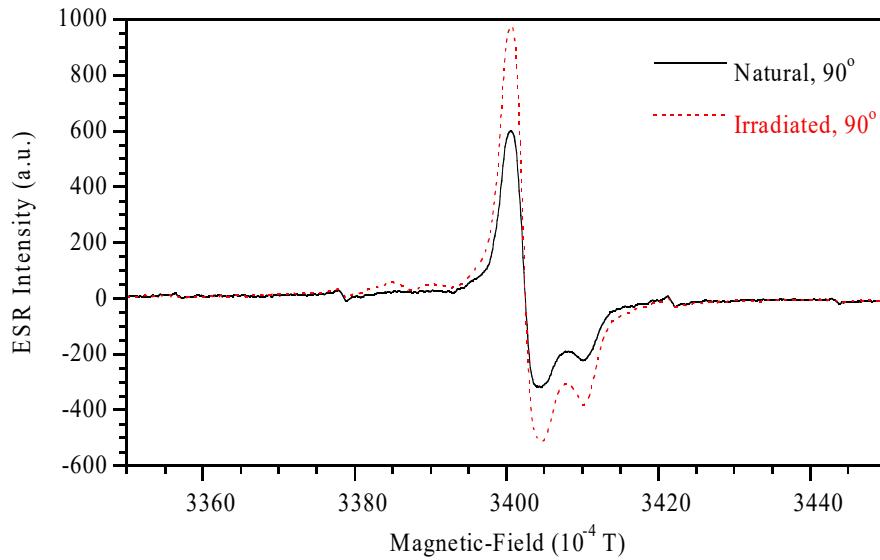
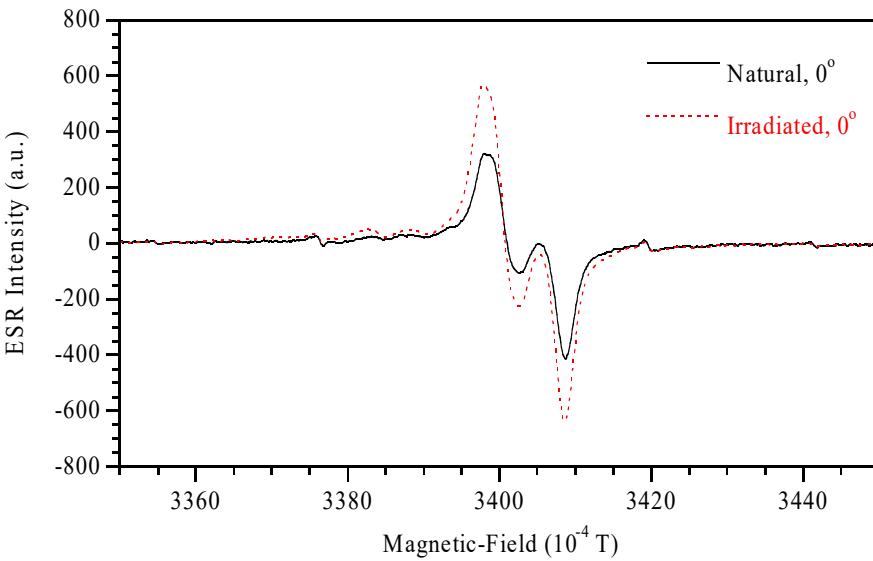
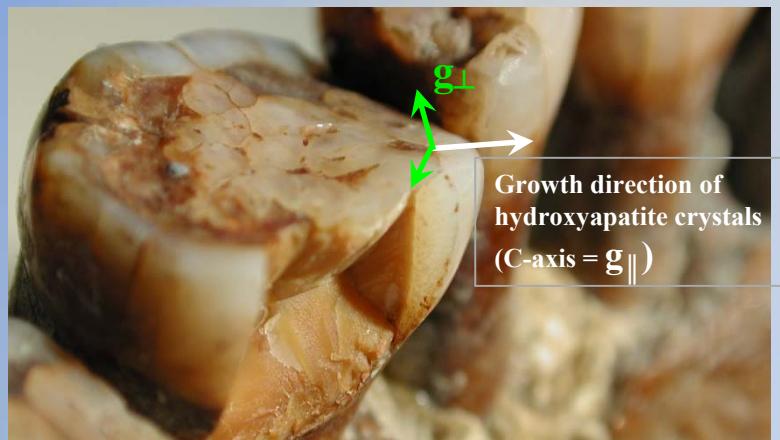


	p enamel	p dentine	p cementum	Age (ka)
Am0007	-0.59	-0.78	-1*	366 +53 / -51
Am0006	-0.53	-0.84	-1*	314 +48 / -45
Am0003	17	25	25	316 ± 26
Am0002	12	20	20	284 ± 17
Am0001	24	27	27	286 ± 29

**ESR-U-Th dating of human remains
(Grün, 2006)**



Direct dating Tabun mandible important in the frame of burial in order to check with TL dates performed on quartz extracted from the layers..





BHS



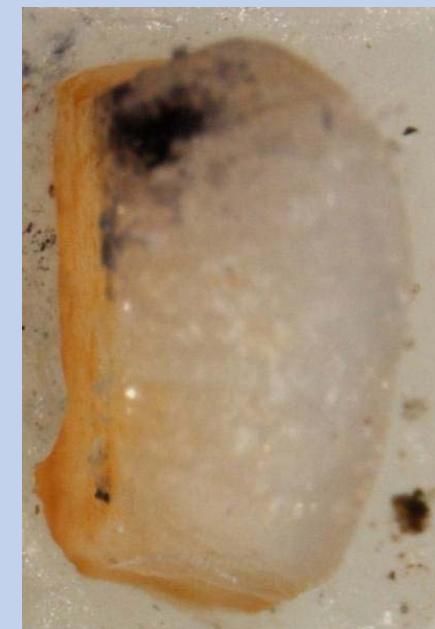
BHL

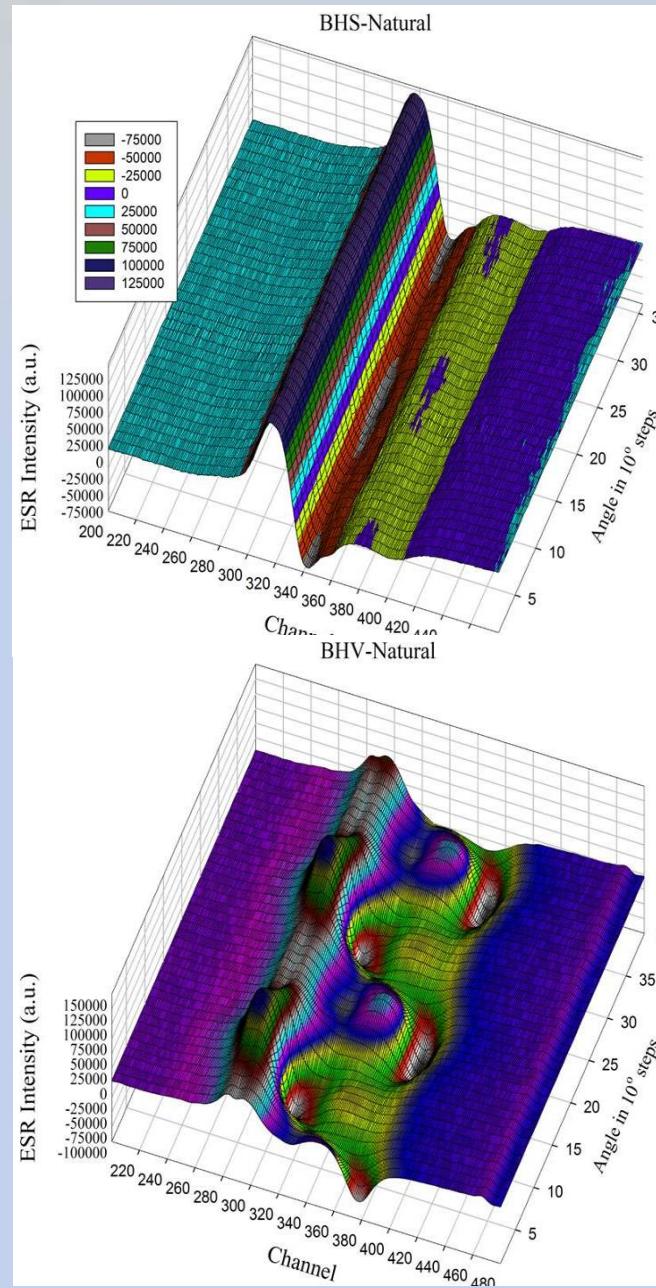
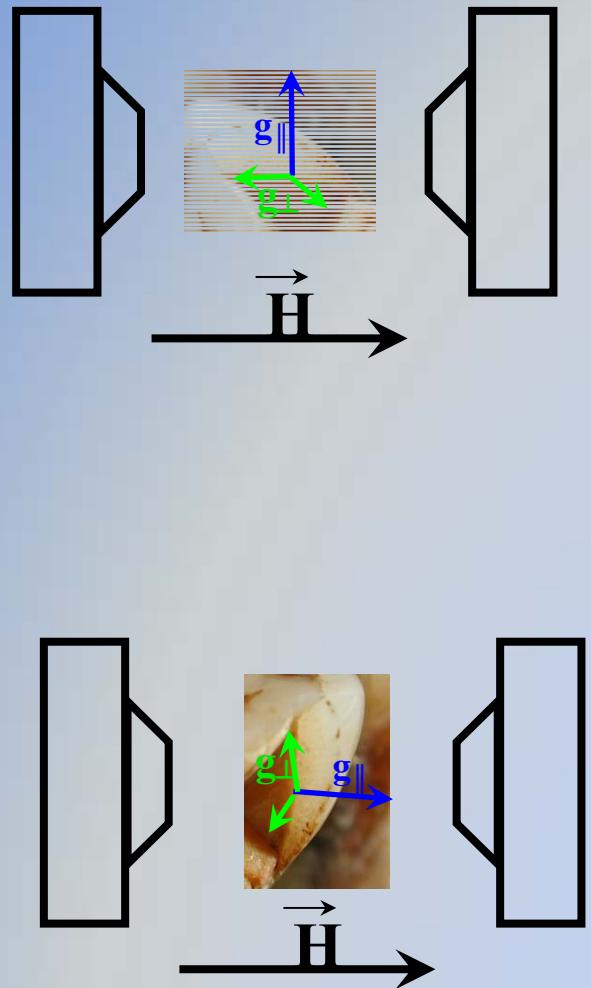


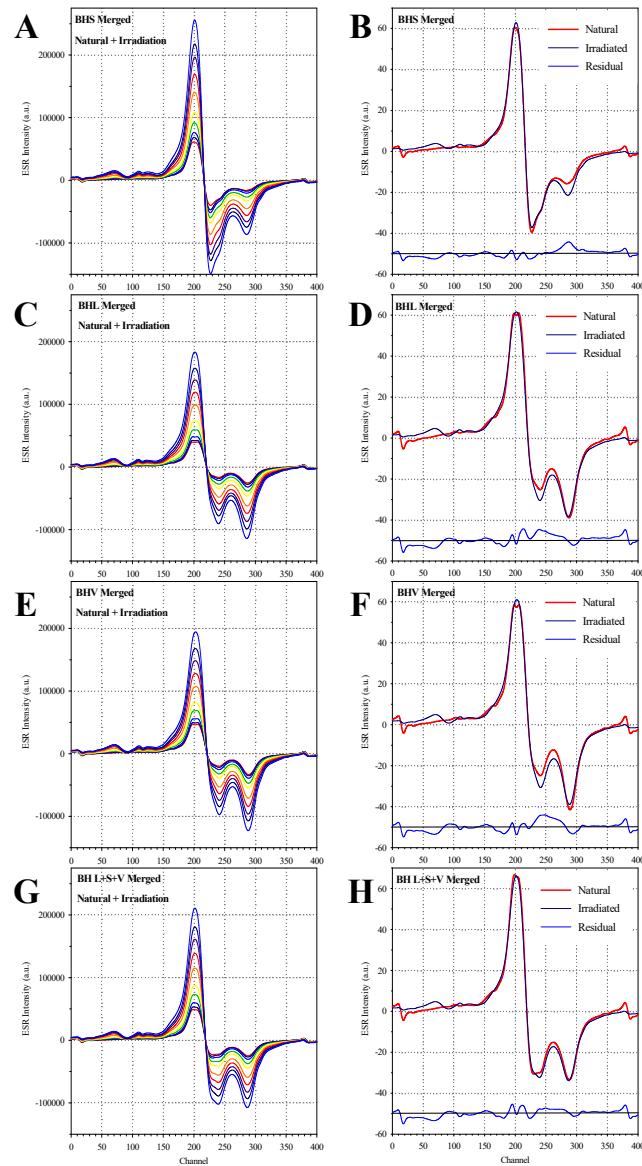
BHV



80/43







- When the spectra are merged, it behaves like a powder (mainly CO₂-radicals):
- There are two different CO₂-radicals, one orientated (**AICORs**), one non-orientated (**NOCORs**) giving rise of a powder spectrum at all angles).
- The thermal stability of the non-orientated CO₂-radical is significantly less than the orientated.

